

Accelerated Corrosion Test of a Scratched Galvanized Steel Sample

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Introduction

An established method for benchmarking the atmospheric corrosion resistance of materials is the use of accelerated corrosion tests (ACT) performed in climate chambers. The main purpose of ACTs is to make reliable and fast predictions of the long-time behavior. The tests expose samples to cyclic conditions during a few weeks. Elevated temperature, repeated drying and wetting together with salt addition are conditions that typically speed up corrosion and characterize ACTs. The samples can be of all shapes and sizes, in setups targeting crevice or galvanic corrosion, or have artificial damage.

This example studies a galvanized steel sample with crossing scratches that fully penetrate the zinc coating and expose the underlying steel. The corrosion is simulated for a dummy ACT running for 7 days. The model solves for a thin liquid film that covers the sample surface. Local variations in pH, corrosion products, and coating damage are shown.



Model Definition



Figure 1: The galvanized steel sample geometry with 0.1 mm wide scratches exposing the steel. The model geometry is marked with dashed lines.

As indicated in the figure, due to symmetry, only one quarter of the sample is needed to investigate the full sample surface. The 2D geometry neglects the thickness of the liquid film. Since the atmospheric corrosion is limited to thin films, in the range of up to tens of micrometers, negligible gradients across the film thickness are expected and makes the thickness dimension redundant.

More information on atmospheric corrosion can be found in the Atmospheric Corrosion example.

ACCELERATED CORROSION TEST MODEL

The ACT is displayed in Figure 2. The temperature interval is between 278.15 K and 323.15 K, and the relative humidity ranges between 70% and 95%. The low RH periods are oscillating around the deliquescence of the NaCl salt (~RH 75%). 1 wt% NaCl solution is sprayed onto the sample at the beginning of day 1, 3, and 6.



Figure 2: ACT. Solid black line indicates periods of spraying 1 wt% NaCl solution onto the sample.

During spray periods, the liquid film thickness is assumed to be constant at $100 \mu m$ and the film volume is fully replenished ten times. When not sprayed, the thickness depends on both the salt load density and the relative humidity. Low RH dries up the film while

higher RH leads to condensation of gaseous water which thickens the film. Below the deliquescence RH, the film is assumed to be discontinuous.

The dependence of RH on film thickness and (other properties) is described in the Atmospheric Corrosion example. A general approximation from that example and throughout this model is that parameters that possibly could depend on the total aqueous species concentration in the film (or ionic strength) are dependent on the NaCl concentration only. The weak temperature dependence that is characteristic for the NaCl salt solubility and deliquescence is also practiced.

Mass transport of several relevant species, reactions (electrochemical, homogeneous, and heterogeneous) and interactions with atmosphere (gas dissolution and drying/ condensation) are all phenomena considered during the ACT. For simplicity a horizontal sample orientation is considered.

ELECTROPHORETIC TRANSPORT INTERFACE

The interface is used to define material balances accounting for the mass transport of species, i, and various sources in a liquid film covering the galvanized steel sample:

$$\frac{\partial c_i}{\partial t} + \nabla \cdot J_i = S \tag{1}$$

In the equation, J_i is the diffusion and migration flux, and S any type of source (mol/(m³·s)).

No conditions can be defined at atmosphere-liquid and liquid-metallic surface boundaries using the interface for the 2D model geometry. However, since the film is thin with no gradients along the film thickness all boundary conditions are converted to sources. For instance for a flux, N_m , the source, S_m , is given by

$$S_m = \frac{N_m}{L} \tag{2}$$

where *L* is the thickness of the liquid film.

Electrochemical Reactions

The electrochemical reactions on zinc and steel are defined using polarization data available in the **Corrosion** folder in the **Material Library**. Metal dissolution is set at zinc and oxygen reduction on steel. Thus these two reactions are accounted for:

$$Zn(s) \rightarrow Zn^{2+} + 2e^{-1}$$
$$\frac{1}{2}O_2 + H_2O + 2e^{-1} \rightarrow 2OH$$

The metal dissolution data are adjusted for changes in salt concentration using a simplified linear dependence and the oxygen reduction reaction uses a simple linear dependence approximation of the dissolved oxygen concentration. Both reactions include an Arrhenius equation factor to incorporate the temperature fluctuations. The reactions are added to as volumetric currents (A/m³), $I_{v,Ox}$ and $I_{v,Red}$, using **Current Source** nodes. The added formulations are

$$I_{\rm v,Ox} = \frac{1-\theta}{L} \cdot \frac{c_{\rm NaCl}}{c_{\rm NaCl,ref}} \cdot i_{\rm loc,Zn} \cdot e^{\frac{E_a}{R_g} \left(\frac{1}{T_{\rm ref}} - \frac{1}{T}\right)}$$
(3)

$$I_{\rm v,Red} = \frac{1-\theta}{L} \cdot \frac{c_{\rm O_2,sat}}{c_{\rm O_2,sat,ref}} \cdot i_{\rm loc, Fe} \cdot e^{\frac{E_a}{R_g} \left(\frac{1}{T_{\rm ref}} - \frac{1}{T}\right)}$$
(4)

where $i_{\rm loc,Zn}$ and $i_{\rm loc,Fe}$ are the polarization data (current density versus potential) for the two reactions, θ the corrosion product surface coverage degree, E_a the activation energy, and subscript "ref" indicate parameter values at experimental data conditions. The film discontinuity below RH deliquescence is accounted for in the reaction sources, by turning $S_{\rm Ox}$ off using a **Ramp** function.

The flux of zinc ions into the film from the metal dissolution is accounted for using a **Species Source** node added under the **Ampholyte** node for aqueous zinc species (see next section). The flux is defined as a source, $S_{Ox,Zn}$, given by

$$S_{\text{Ox},Zn} = \frac{I_{\text{v},\text{Ox}}}{2F} \tag{5}$$

Homogeneous Reactions in Liquid Film

Carbonates originating from the atmosphere are accounted for. Two deprotonation steps are defined, starting with the deprotonation of carbonic acid equilibrium reaction:

$$H_{2}CO_{3} \Leftrightarrow H^{+} + HCO_{3}^{-}$$
$$HCO_{3}^{-} \Leftrightarrow H^{+} + CO_{3}^{-}$$

Hydrolysis of zinc ions from the dissolution of the zinc coating are considered in four steps (Ref. 2):

$$Zn^{2+} + H_2O \Leftrightarrow Zn(OH)^+ + H^+$$
$$Zn(OH)^+ + H_2O \Leftrightarrow Zn(OH)_2(aq) + H^+$$
$$Zn(OH)_2(aq) + H_2O \Leftrightarrow Zn(OH)_3^- + H^+$$
$$Zn(OH)_3^- + H_2O \Leftrightarrow Zn(OH)_4^{2-} + H^+$$

Two **Ampholyte** nodes are used in the model. One node defines the mass balance accounting for all forms of the aqueous zinc species $(Zn^{2+}, Zn(OH)^+, Zn(OH)_2(aq), Zn(OH)_3^-$, and $Zn(OH)_4^{2-}$) reactions and the other does the same for the carbonate $(H_2CO_3, HCO_3^-, and CO_3^{2-})$ reactions.

NaCl Salt Species

The film conductivity together with the salt dependent parameters requires the presence of sodium and chloride ions. Both ions are added to using the **Fully Dissociated Species** node.

Corrosion Products and Passivation

Corrosion products are formed can passivate the metallic surface. This example accounts for the formation of ZnO in the following reaction:

$$\operatorname{Zn}^{2+} + 2\operatorname{OH}^{-} \Leftrightarrow \operatorname{ZnO}(s) + \operatorname{H}_{2}\operatorname{O}$$

The reaction is defined as fully reversible (ZnO both precipitates and dissolves). It is added as a reaction source, S_{Zn} , in a **Species Source** node in the **Ampholyte** node for the zinc species. The source formulation is based on the corrosion product solubility, as follows:

$$S_{\rm Zn} = -\frac{k_{\rm ZnO}}{L} (c_{\rm Zn^{2+}} c_{\rm OH^-}^2 - K_{\rm S,ZnO}), \text{ if prec.; } (c_{\rm Zn^{2+}} c_{\rm OH^-}^2 - K_{\rm SP,ZnO}) > 0$$
(6)

$$S_{\rm Zn} = -\theta \frac{k_{\rm ZnO}}{L} (c_{\rm Zn^{2+}} c_{\rm OH^-}^2 - K_{\rm S,ZnO}), \text{ if diss.; } (c_{\rm Zn^{2+}} c_{\rm OH^-}^2 - K_{\rm SP,ZnO}) < 0$$
(7)

In the above, k_{ZnO} is the rate constant for the corrosion product conversion (mol/(m²·s)) and $K_{S,ZnO}$ the equilibrium constant of the reaction.

The precipitated ZnO is set to cover the sample surface and assumed to inhibit both electrochemical reactions (see above). The ZnO coverage degree, θ , is assumed to change according to the following expression (Ref. 3):

$$\theta = 1 - e^{-\frac{m_{ZnO}}{m_{\text{tot,surf}}}}$$
(8)

where m_{ZnO} is the precipitated molar amount of ZnO per surface area (mol/m²) and $m_{tot,surf}$ is the molar metal surface availability per surface area for ZnO precipitation.

The precipitated ZnO amount is computed locally on the surface using a **Coefficient Form PDE**.

Dissolved Atmospheric Gases

Two atmospheric gases are accounted for.

Carbon dioxide dissolves into the film and affects the carbonate concentration and pH. The aqueous carbon dioxide saturation concentration, $c_{\rm CO2,sat}$, depends on the partial pressure, $p_{\rm CO2}$, together with temperature and the salt concentration (Ref. 1). A reaction source, $S_{\rm carbonate}$, is defined in a **Species Source** node in the **Ampholyte** node for the carbonte species. The source minimizes the difference between the carbonate concentration in the film and the saturation concentration, as follows:

$$S_{\text{carbonate}} = \varepsilon \frac{k_{\text{CO}_2,\text{sat}}}{L} (c_{\text{CO}_2,\text{sat}}(p_{\text{CO}_2}, T, c_{\text{NaCl}}) - c_{\text{H}_2\text{CO}_3}), \quad c_{\text{CO}_2(\text{aq})} \approx c_{\text{H}_2\text{CO}_3} \quad (9)$$

In the expression, $k_{\text{CO2,sat}}$ is the rate constant for the carbon dioxide (1/s). The expression assumes that the carbonic acid concentration is mainly in the form of dissolved carbon dioxide in the liquid, as the following equilibrium reaction for carbon dioxide in liquid is strongly driven toward the left (Ref. 3):

$$CO_2(aq) + H_2O \Leftrightarrow H_2CO_3$$

Oxygen dissolves into the film as well which affects the oxygen reduction reaction. The dissolved oxygen concentration is equal to the saturation concentration that is dependent on the oxygen content in the atmosphere, temperature and salt concentration (compare with Atmospheric Corrosion).

Drying and Condensation

The liquid film thickness varies with drying/evaporation that in turn affects species concentration. Therefore, an expanded formulation of the accumulation term in the material balances needs to be defined, as follows:

$$\frac{1}{L} \cdot \frac{\partial (c_i L)}{\partial t} + \nabla \cdot J_i = S$$
(10)

or

$$\frac{\partial c_i}{\partial t} + \frac{\varepsilon c_i}{L} \cdot \frac{\partial L}{\partial t} + \nabla \cdot J_i = S$$
(11)

The second term on the left-hand side in the equation is defined in **Species Source** nodes for the species.

Spraying

During spraying, aqueous species concentrations in the liquid are replaced with the compositions of the 1 wt% NaCl spray solution, $c_{i,spray}$. This is defined using **Species Source** nodes for the species. The generalized source expression used is as follows:

$$S_{\text{NaCl}} = k_{\text{spray}}(c_{i, \text{spray}} - c_i)$$
(12)

The rate constant, k_{sprav} (1/s), controls how well the solution is replenished.

Results and Discussion

Several variables are monitored during the run giving indications of the aggressiveness of the ACT (Figure 3, Figure 4, and Figure 5). In Figure 6 and Figure 7, the corrosion damage is shown in terms of loss of zinc coating mass and maximum decrease in coating thickness. At low RH the corrosion progresses very slowly. The precipitated amount of





Figure 3: Salt concentration and liquid film thickness during ACT.



Figure 4: Total current of electrochemical reactions during ACT.



Figure 5: Maximum and minimum pH at sample during ACT.







Figure 7: Maximum thickness decrease of coating during ACT.





Figure 8: Local coating thickness decrease after 7 days.



The localized pH is displayed in Figure 9 for different times. The oxygen reduction reaction in the scratch keeps the pH basic there at all times, favoring passivation.

pH on sample at different times

Figure 9: Local pH at the sample surface at different times during the ACT.



Figure 10: Local degree of coverage after 7 days.

References

1. M. Nordsveen, S. Nesic, R. Nyborg, and A. Stangeland, "A Mechanistic Model for Carbon Dioxide Corrosion of Mild Steel in the Presence of Protective Iron Carbonate Films-Part 1: Theory and Verification," *Corrosion*, vol. 59, no. 5, pp. 443–455, 2023

2. V. Topa, A.S. Demeter, L. Hotoiu, D. Deconinck, and J. Deconinck, "A transient multiion transport model for galvanized steel corrosion protection," *Electrochimica Acta*, vol. 77, pp. 339–347, 2012.

3. T.G. Zavalis, M. Ström, D. Persson, E. Wendel, J. Ahlström, K.B. Törne, C. Taxén, B. Rendahl, J. Voltaire, K. Eriksson, D. Thierry, and J. Tidblad, "Mechanistic Model with Empirical Pitting Onset Approach for Detailed and Efficient Virtual Analysis of Atmospheric Bimetallic Corrosion," *Materials*, vol. 16, pp. 923–946, 2023.

Application Library path: Corrosion_Module/Atmospheric_Corrosion/ act_scratched_galvanized_steel

Modeling Instructions

From the File menu, choose New.

NEW

In the New window, click 🔗 Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click 9 2D.
- 2 In the Select Physics tree, select Chemical Species Transport > Electrophoretic Transport (el).
- 3 Click Add.
- 4 Click 🔿 Study.
- 5 In the Select Study tree, select Preset Studies for Selected Physics Interfaces > Time Dependent with Initialization.
- 6 Click M Done.

GLOBAL DEFINITIONS

Load the model parameters from a text file.

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** Click **b** Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_parameters.txt.

GEOMETRY I

Draw the geometry.

Rectangle 1 (r1)

- I In the Geometry toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type w_sample.
- 4 In the **Height** text field, type h_sample.

Rectangle 2 (r2)

I In the **Geometry** toolbar, click **Rectangle**.

- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type w_sample.
- 4 In the **Height** text field, type h_scratch.
- **5** Locate the **Position** section. From the **Base** list, choose **Center**.
- 6 In the x text field, type w_sample/2.
- 7 In the y text field, type h_sample/2.
- 8 Locate the Rotation Angle section. In the Rotation text field, type 45.

Rectangle 3 (r3)

- I In the **Geometry** toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type w_sample.
- **4** In the **Height** text field, type h_scratch.
- **5** Locate the **Position** section. From the **Base** list, choose **Center**.
- 6 In the x text field, type w_sample/2.
- 7 In the y text field, type h_sample/2.
- 8 Locate the Rotation Angle section. In the Rotation text field, type 315.

9 Click 🟢 Build All Objects.

Your geometry should now look like this:



Due to symmetry, one quarter of the drawn geometry is modeled. The upper right corner is selected as model geometry.

Rectangle 4 (r4)

- I In the **Geometry** toolbar, click **Rectangle**.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type w_sample/2.
- 4 In the **Height** text field, type h_sample.

Rectangle 5 (r5)

- I In the Geometry toolbar, click 📃 Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type w_sample/2.
- 4 In the **Height** text field, type h_sample/2.
- 5 Locate the **Position** section. In the **x** text field, type w_sample/2.

Difference I (dif1)

I In the Geometry toolbar, click 🔲 Booleans and Partitions and choose Difference.

- 2 Select the objects r1, r2, and r3 only.
- 3 In the Settings window for Difference, locate the Difference section.
- **4** Click to select the **Delta Activate Selection** toggle button for **Objects to subtract**.
- 5 Select the objects r4 and r5 only.
- 6 Click 🟢 Build All Objects.
- 7 Click the 🕂 Zoom Extents button in the Graphics toolbar.

Your model geometry should now look like this:



Make selections to facilitate the model setup.

DEFINITIONS

Zinc

- I In the **Definitions** toolbar, click **here Explicit**.
- 2 In the Settings window for Explicit, type Zinc in the Label text field.
- **3** Select Domain 2 only.

Steel

- I In the **Definitions** toolbar, click **here Explicit**.
- 2 In the Settings window for Explicit, type Steel in the Label text field.

3 Select Domains 1 and 3 only.

Wetted Surface

- I In the **Definitions** toolbar, click http://www.explicit.
- 2 In the Settings window for Explicit, type Wetted Surface in the Label text field.
- 3 Locate the Input Entities section. Select the All domains checkbox.

Add some integration operators for probes and postprocessing.

Integration - Zn

- I In the Definitions toolbar, click 🖉 Nonlocal Couplings and choose Integration.
- 2 In the Settings window for Integration, type intop_zinc in the Operator name text field.
- 3 In the Label text field, type Integration Zn.
- 4 Locate the Source Selection section. From the Selection list, choose Zinc.

Integration - Steel

- I In the Definitions toolbar, click *P* Nonlocal Couplings and choose Integration.
- **2** In the **Settings** window for **Integration**, type **Integration Steel** in the **Label** text field.
- 3 In the **Operator name** text field, type intop_steel.
- 4 Locate the Source Selection section. From the Selection list, choose Steel.

Integration - Wetted Surface

- I In the Definitions toolbar, click *N* Nonlocal Couplings and choose Integration.
- 2 In the Settings window for Integration, type Integration Wetted Surface in the Label text field.
- 3 In the **Operator name** text field, type intop_wet.
- 4 Locate the Source Selection section. From the Selection list, choose Wetted Surface.

The ACT cycle is added as three separate interpolation files. The first describes the variation in RH, the second temperature, and the third spraying periods over time. Use the piecewise cubic interpolation alternative for smoother transitions.

Interpolation - Relative Humidity

- I In the **Definitions** toolbar, click \bigwedge **Interpolation**.
- **2** In the **Settings** window for **Interpolation**, type Interpolation Relative Humidity in the **Label** text field.
- 3 Locate the Definition section. In the Function name text field, type RH_ACT.

- 4 Click **by Load from File**.
- 5 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_rh.txt.
- 6 Locate the Interpolation and Extrapolation section. From the Interpolation list, choose Piecewise cubic.
- 7 Locate the Units section. In the Function table, enter the following settings:

Function	Unit
RH_ACT	1

8 In the Argument table, enter the following settings:

Argument	Unit
t	S

Interpolation - Temperature

- I In the **Definitions** toolbar, click \bigwedge **Interpolation**.
- 2 In the Settings window for Interpolation, type Interpolation Temperature in the Label text field.
- **3** Locate the **Definition** section. In the **Function name** text field, type T_ACT.
- 4 Click 📂 Load from File.
- 5 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_temperature.txt.
- 6 Locate the Interpolation and Extrapolation section. From the Interpolation list, choose Piecewise cubic.
- 7 Locate the Units section. In the Function table, enter the following settings:

Function	Unit
T_ACT	К

8 In the Argument table, enter the following settings:

Argument	Unit
t	S

Interpolation - Spray

I In the **Definitions** toolbar, click \bigwedge **Interpolation**.

- **2** In the **Settings** window for **Interpolation**, type **Interpolation Spray** in the **Label** text field.
- 3 Locate the Definition section. In the Function name text field, type spray_ACT.
- 4 Click 📂 Load from File.
- 5 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_spray.txt.
- 6 Locate the Interpolation and Extrapolation section. From the Interpolation list, choose Piecewise cubic.
- 7 Locate the Units section. In the Function table, enter the following settings:

Function	Unit
spray_ACT	1

8 In the Argument table, enter the following settings:

Argument	Unit
t	S

Add zinc and iron from the Corrosion branch in the Material Library.

ADD MATERIAL

- I In the Materials toolbar, click 🙀 Add Material to open the Add Material window.
- 2 Go to the Add Material window.
- 3 In the tree, select Corrosion > Elements > Fe in 3% NaCl.
- 4 Click the Add to Component button in the window toolbar.
- 5 In the tree, select Corrosion > Elements > Zn in aerated 3.5 wt% NaCl.
- 6 Click the Add to Component button in the window toolbar.
- 7 In the Materials toolbar, click 🙀 Add Material to close the Add Material window.

MATERIALS

Fe in 3% NaCl (mat1)

- I In the Settings window for Material, locate the Geometric Entity Selection section.
- 2 From the Selection list, choose Steel.

Zn in aerated 3.5 wt% NaCl (mat2)

- I In the Model Builder window, click Zn in aerated 3.5 wt% NaCl (mat2).
- 2 In the Settings window for Material, locate the Geometric Entity Selection section.

3 From the **Selection** list, choose **Zinc**.

Add variables describing the system. Unknown variables warnings will be resolved as soon as the physics have been set up.

DEFINITIONS

Variables - Global

- I In the **Definitions** toolbar, click **a**= **Local Variables**.
- 2 In the Settings window for Variables, type Variables Global in the Label text field.
- **3** Locate the Variables section. Click *b* Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_global_variables.txt.

Variables - Rates

- I In the **Definitions** toolbar, click $\partial =$ **Local Variables**.
- 2 In the Settings window for Variables, type Variables Rates in the Label text field.
- **3** Locate the Geometric Entity Selection section. From the Geometric entity level list, choose Domain.
- 4 From the Selection list, choose Wetted Surface.
- 5 Locate the Variables section. Click 📂 Load from File.
- 6 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_rate_variables.txt.

Variables - Film

- I In the **Definitions** toolbar, click $\partial =$ **Local Variables**.
- 2 In the Settings window for Variables, type Variables Film in the Label text field.
- **3** Locate the **Geometric Entity Selection** section. From the **Geometric entity level** list, choose **Domain**.
- 4 From the Selection list, choose Wetted Surface.
- 5 Locate the Variables section. Click 📂 Load from File.
- 6 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_film_variables.txt.

Variables - Zn

- I In the **Definitions** toolbar, click $\partial =$ Local Variables.
- 2 In the Settings window for Variables, locate the Geometric Entity Selection section.

- 3 From the Geometric entity level list, choose Domain.
- 4 From the Selection list, choose Zinc.
- 5 In the Label text field, type Variables Zn.
- 6 Locate the Variables section. Click 📂 Load from File.
- 7 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_zn_variables.txt.

Variables - Steel

- I In the **Definitions** toolbar, click **a**= **Local Variables**.
- 2 In the Settings window for Variables, locate the Geometric Entity Selection section.
- 3 From the Geometric entity level list, choose Domain.
- 4 From the Selection list, choose Steel.
- 5 In the Label text field, type Variables Steel.
- 6 Locate the Variables section. Click 📂 Load from File.
- 7 Browse to the model's Application Libraries folder and double-click the file act_scratched_galvanized_steel_fe_variables.txt.

Start defining the transport and reactions in the **Electrophoretic Transport** interface that was added.

ELECTROPHORETIC TRANSPORT (EL)

- I In the Model Builder window, under Component I (compl) click Electrophoretic Transport (el).
- **2** In the **Settings** window for **Electrophoretic Transport**, locate the **Out-of-Plane Thickness** section.
- **3** In the d_z text field, type d_film.

Solvent I

- I In the Model Builder window, under Component I (compl) > Electrophoretic Transport (el) click Solvent I.
- 2 In the Settings window for Solvent, locate the Model Input section.
- **3** From the *T* list, choose **User defined**. In the associated text field, type $T_ACT(t)$.
- 4 Locate the Diffusion and Migration section. From the Specify list, choose Diffusivity.
- **5** In the $D_{\rm H}$ text field, type DH.
- 6 In the $D_{\rm OH}$ text field, type DOH.

Initial Potential 1

- I In the Model Builder window, click Initial Potential I.
- 2 In the Settings window for Initial Potential, locate the Initial Potential section.
- **3** In the *phil* text field, type V0.

Use an **Ampholyte** node to model the hydrolysis equilibrium reactions of the aqueous zinc species.

Ampholyte - Hydrolysis Zn species

- I In the Physics toolbar, click **Domains** and choose Ampholyte.
- 2 In the Settings window for Ampholyte, type Ampholyte Hydrolysis Zn species in the Label text field.
- **3** Locate the **Ampholyte** section. In the **Species name** text field, type Zn.
- **4** In the *k* text field, type 4.
- **5** In the table, enter the following settings:

Dissociation step (1)	pKa (I)
I	-log10(K_Znstep1)
2	-log10(K_Znstep2)
3	-log10(K_Znstep3)
4	-log10(K_Znstep4)

- **6** In the z_0 text field, type -2.
- 7 Locate the Diffusion and Migration section. From the Specify list, choose Diffusivity.
- **8** In the D text field, type DZn.

The **Species Source** subnodes are available once advanced physics options are enabled. Use these to add corrosion product precipitation and spraying together with drying and condensation of the liquid film.

- 9 Click the 🐱 Show More Options button in the Model Builder toolbar.
- **10** In the **Show More Options** dialog, in the tree, select the checkbox for the node **Physics** > **Advanced Physics Options**.
- II Click OK.

Species Source - ZnO Corrosion Product

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source ZnO Corrosion Product in the Label text field.

3 Locate the **Species Source** section. In the S text field, type $-r_Zn0$.

Ampholyte - Hydrolysis Zn species

In the Model Builder window, click Ampholyte - Hydrolysis Zn species.

Species Source - Zn Dissolution

- I In the Physics toolbar, click 📻 Attributes and choose Species Source.
- 2 In the Settings window for Species Source, locate the Domain Selection section.
- **3** From the **Selection** list, choose **Zinc**.
- 4 Locate the Species Source section. In the S text field, type N_0x .
- 5 In the Label text field, type Species Source Zn Dissolution.

Ampholyte - Hydrolysis Zn species

In the Model Builder window, click Ampholyte - Hydrolysis Zn species.

Species Source - Zn Removal with Spraying

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Zn Removal with Spraying in the Label text field.
- **3** Locate the **Species Source** section. In the *S* text field, type rSpray_Zn.

Ampholyte - Hydrolysis Zn species

In the Model Builder window, click Ampholyte - Hydrolysis Zn species.

Species Source - Drying and Condensation

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Drying and Condensation in the Label text field.
- 3 Locate the Species Source section. In the S text field, type rDry_Zn.

Use an additional **Ampholyte** node to model the carbonate deprotonation equilibrium reactions.

Ampholyte - Deprotonation Carbonate Species

- I In the Physics toolbar, click **Domains** and choose Ampholyte.
- 2 In the Settings window for Ampholyte, type Ampholyte Deprotonation Carbonate Species in the Label text field.
- 3 Locate the Ampholyte section. In the Species name text field, type CO3.

4 In the table, enter the following settings:

Dissociation step (1)	pKa (I)
I	-log10(K_CO3step1)
2	-log10(K_CO3step2)

- **5** In the z_0 text field, type -2.
- 6 Locate the Diffusion and Migration section. From the Diffusivity-mobility settings list, choose Individual.
- 7 In the table, enter the following settings:

Index (I)	Charge number (I)	Diffusivity (m^2/s)	Mobility (s*mol/kg)
0	-2	DC03	el.D1_CO3/ R_const/T_ACT(t)
I	-1	DHC03	el.D2_CO3/ R_const/T_ACT(t)
2	0	DH2C03	el.D3_CO3/ R_const/T_ACT(t)

Species Source - Carbonates from Atmosphere

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Carbonates from Atmosphere in the Label text field.
- **3** Locate the **Species Source** section. In the *S* text field, type r_C03.

Ampholyte - Deprotonation Carbonate Species

In the Model Builder window, click Ampholyte - Deprotonation Carbonate Species.

Species Source - Carbonate Removal with Spraying

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Carbonate Removal with Spraying in the Label text field.
- 3 Locate the Species Source section. In the S text field, type rSpray_C03.

Ampholyte - Deprotonation Carbonate Species

In the Model Builder window, click Ampholyte - Deprotonation Carbonate Species.

Species Source - Drying and Condensation

I In the Physics toolbar, click — Attributes and choose Species Source.

- 2 In the Settings window for Species Source, type Species Source Drying and Condensation in the Label text field.
- **3** Locate the **Species Source** section. In the *S* text field, type rDry_C03.

Initial Concentration 1

- I In the Model Builder window, click Initial Concentration I.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the c text field, type cC030.

Add the NaCl as two fully dissociated species (sodium and chloride ions) with the **Fully Dissociated Species** node.

Fully Dissociated Species - Na

- I In the Physics toolbar, click 🔵 Domains and choose Fully Dissociated Species.
- In the Settings window for Fully Dissociated Species, type Fully Dissociated Species
 Na in the Label text field.
- 3 Locate the Fully Dissociated Species section. In the Species name text field, type Na.
- **4** In the *z* text field, type 1.
- 5 Locate the Diffusion and Migration section. From the Specify list, choose Diffusivity.
- 6 In the *D* text field, type DNa.

Species Source - Spray

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Spray in the Label text field.
- **3** Locate the **Species Source** section. In the *S* text field, type rSpray_Na.

Fully Dissociated Species - Na

In the Model Builder window, click Fully Dissociated Species - Na.

Species Source - Drying and Condensation

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Drying and Condensation in the Label text field.
- 3 Locate the Species Source section. In the S text field, type rDry_Na.

Initial Concentration 1

- I In the Model Builder window, click Initial Concentration I.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.

3 In the *c* text field, type cNaCl0.

Fully Dissociated Species - Cl

- I In the Physics toolbar, click 🔵 Domains and choose Fully Dissociated Species.
- 2 In the Settings window for Fully Dissociated Species, type Fully Dissociated Species
 Cl in the Label text field.
- 3 Locate the Fully Dissociated Species section. In the Species name text field, type C1.
- 4 In the z text field, type -1.
- 5 Locate the Diffusion and Migration section. From the Specify list, choose Diffusivity.
- 6 In the *D* text field, type DC1.

Species Source - Spray

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Spray in the Label text field.
- **3** Locate the **Species Source** section. In the *S* text field, type rSpray_C1.

Fully Dissociated Species - Cl

In the Model Builder window, click Fully Dissociated Species - Cl.

Species Source - Drying and Condensation

- I In the Physics toolbar, click Attributes and choose Species Source.
- 2 In the Settings window for Species Source, type Species Source Drying and Condensation in the Label text field.
- 3 Locate the Species Source section. In the S text field, type rDry_C1.

Current Source 1

In the **Physics** toolbar, click **Domains** and choose **Current Source**.

Initial Concentration 1

- I In the Model Builder window, under Component I (compl) > Electrophoretic Transport (el) > Fully Dissociated Species - Cl click Initial Concentration I.
- 2 In the Settings window for Initial Concentration, locate the Initial Concentration section.
- **3** In the *c* text field, type cNaCl0.

Local electrochemical reactions are added using **Current Source** nodes. The reactions are defined in the imported Zn and steel variables.

Current Source - Oxygen Reduction

- I In the Model Builder window, under Component I (compl) > Electrophoretic Transport (el) click Current Source I.
- 2 In the Settings window for Current Source, type Current Source Oxygen Reduction in the Label text field.
- 3 Locate the Domain Selection section. From the Selection list, choose Steel.
- 4 Locate the Electrolyte Current Source section. In the Q_1 text field, type Q_red.

Current Source - Zn Metal Dissolution

- I In the Physics toolbar, click **Domains** and choose Current Source.
- 2 In the Settings window for Current Source, type Current Source Zn Metal Dissolution in the Label text field.
- 3 Locate the Domain Selection section. From the Selection list, choose Zinc.
- **4** Locate the **Electrolyte Current Source** section. In the Q_1 text field, type Q_ox.

Add a **Coefficient Form PDE** interface and define dissolving and depositing species; zinc metal dissolution and ZnO dissolution/deposition.

ADD PHYSICS

- I In the Physics toolbar, click 🙀 Add Physics to open the Add Physics window.
- 2 Go to the Add Physics window.
- 3 In the tree, select Mathematics > PDE Interfaces > Coefficient Form PDE (c).
- 4 Click the Add to Component I button in the window toolbar.
- 5 In the Physics toolbar, click 🙀 Add Physics to close the Add Physics window.

DISSOLVING AND DEPOSITING SPECIES

- I In the Settings window for Coefficient Form PDE, type Dissolving and Depositing Species in the Label text field.
- 2 Locate the Units section. Click **Select Dependent Variable Quantity**.
- 3 In the Physical Quantity dialog, select Transport > Surface site concentration (mol/m²) in the tree.
- 4 Click OK.
- 5 In the Settings window for Coefficient Form PDE, locate the Units section.
- 6 Click Select Source Term Quantity.
- 7 In the Physical Quantity dialog, select Transport > Molar flux (mol/(m^2*s)) in the tree.
- 8 Click OK.

9 In the **Settings** window for **Coefficient Form PDE**, click to expand the **Dependent Variables** section.

10 In the Dependent variables (mol/m²) table, enter the following settings:

mZn

II Click + Add Dependent Variable.

12 In the Dependent variables (mol/m²) table, enter the following settings:

mZn

mZnO

Zn

- I In the Model Builder window, under Component I (compl) > Dissolving and Depositing Species (c) click Coefficient Form PDE I.
- 2 In the Settings window for Coefficient Form PDE, type Zn in the Label text field.
- **3** Locate the **Diffusion Coefficient** section. In the *c* text-field array, type 0 in the first column of the first row.
- 4 In the *c* text-field array, type 0 in the second column of the second row.
- 5 Locate the Source Term section. In the *f* text-field array, type N_0x*d_film on the first row.
- 6 In the *f* text-field array, type r_ZnO*d_film on the second row.

Steel

- I In the Physics toolbar, click 🔵 Domains and choose Coefficient Form PDE.
- 2 In the Settings window for Coefficient Form PDE, type Steel in the Label text field.
- **3** Locate the **Domain Selection** section. From the **Selection** list, choose **Steel**.
- **4** Locate the **Diffusion Coefficient** section. In the *c* text-field array, type 0 in the first column of the first row.
- 5 In the *c* text-field array, type 0 in the second column of the second row.
- **6** Locate the **Source Term** section. In the f text-field array, type 0 on the first row.
- 7 In the *f* text-field array, type r_Zn0*d_film on the second row.

DEFINITIONS

Ramp - Deliquescence Limit

I In the Home toolbar, click f(x) Functions and choose Local > Ramp.

- 2 In the Settings window for Ramp, type Ramp Deliquescence Limit in the Label text field.
- **3** In the **Function name** text field, type rm_deliq.
- **4** Locate the **Parameters** section. In the **Location** text field, type ramp_loc.
- **5** In the **Slope** text field, type ramp_s1.
- 6 Select the Cutoff checkbox. In the associated text field, type ramp_co.
- 7 Click to expand the **Smoothing** section.
- 8 Select the Size of transition zone at start checkbox. In the associated text field, type ramp_sm.
- **9** Select the **Size of transition zone at cutoff** checkbox. In the associated text field, type ramp_sm.

Add probes to monitor various behaviors during the ACT simulation and to reduce the amount of output data.

Global Variable Probe - ZnO

- I In the Definitions toolbar, click probes and choose Global Variable Probe.
- 2 In the Settings window for Global Variable Probe, type Global Variable Probe ZnO in the Label text field.
- **3** In the **Variable name** text field, type var_Zn0.
- 4 Locate the Expression section. In the Expression text field, type m_Zn0.
- 5 From the Table and plot unit list, choose mg.
- **6** Select the **Description** checkbox.
- 7 Click to expand the Table and Window Settings section. Click + Add Table.
- 8 Click + Add Plot Window.

Global Variable Probe - Zn Metal

- I In the Definitions toolbar, click probes and choose Global Variable Probe.
- 2 In the Settings window for Global Variable Probe, type Global Variable Probe Zn Metal in the Label text field.
- **3** In the **Variable name** text field, type var_m_Zn.
- 4 Locate the Expression section. In the Expression text field, type m_Zn.
- 5 From the Table and plot unit list, choose mg.
- **6** Select the **Description** checkbox.

- 7 Locate the Table and Window Settings section. From the Output table list, choose Table 1.
- 8 From the Plot window list, choose Probe Plot I.

Global Variable Probe - Total Zinc Metal Dissolution Current

- I In the Definitions toolbar, click probes and choose Global Variable Probe.
- 2 In the Settings window for Global Variable Probe, type Global Variable Probe -Total Zinc Metal Dissolution Current in the Label text field.
- 3 In the Variable name text field, type var_I_ox.
- 4 Locate the Expression section. In the Expression text field, type I_ox.
- **5** From the **Table and plot unit** list, choose μA .
- **6** Select the **Description** checkbox.
- 7 Locate the Table and Window Settings section. Click + Add Table.
- 8 Click + Add Plot Window.

Global Variable Probe - Total Oxygen Reduction Current

- I In the Definitions toolbar, click probes and choose Global Variable Probe.
- 2 In the Settings window for Global Variable Probe, type Global Variable Probe -Total Oxygen Reduction Current in the Label text field.
- 3 In the Variable name text field, type var_I_red.
- 4 Locate the Expression section. In the Expression text field, type I_red.
- 5 From the Table and plot unit list, choose µA.
- 6 Select the **Description** checkbox.
- 7 Locate the Table and Window Settings section. From the Output table list, choose Table 2.
- 8 From the Plot window list, choose Probe Plot 2.

Domain Probe - Maximum Zn Coating Thickness Loss

- I In the Definitions toolbar, click probes and choose Domain Probe.
- 2 In the Settings window for Domain Probe, type Domain Probe Maximum Zn Coating Thickness Loss in the Label text field.
- **3** In the **Variable name** text field, type dom_d_Zn.
- 4 Locate the Probe Type section. From the Type list, choose Maximum.
- 5 Locate the Source Selection section. From the Selection list, choose Zinc.
- 6 Locate the Expression section. In the Expression text field, type d_Zn.

- 7 From the Table and plot unit list, choose µm.
- **8** Select the **Description** checkbox.
- 9 Click to expand the Table and Window Settings section. Click + Add Table.
- **IO** Click + Add Plot Window.

Domain Probe - Maximum pH

- I In the Definitions toolbar, click probes and choose Domain Probe.
- 2 In the Settings window for Domain Probe, type Domain Probe Maximum pH in the Label text field.
- **3** In the **Variable name** text field, type dom_maxpH.
- 4 Locate the Probe Type section. From the Type list, choose Maximum.
- **5** Locate the **Table and Window Settings** section. Click + **Add Table**.
- 6 Click + Add Plot Window.

Domain Probe - Minimum pH

- I Right-click Domain Probe Maximum pH and choose Duplicate.
- 2 In the Settings window for Domain Probe, type Domain Probe Minimum pH in the Label text field.
- 3 In the Variable name text field, type dom_minpH.
- 4 Locate the Probe Type section. From the Type list, choose Minimum.

Domain Probe - Average NaCl Concentration in Liquid Film

- I In the Definitions toolbar, click probes and choose Domain Probe.
- 2 In the Settings window for Domain Probe, type Domain Probe Average NaCl Concentration in Liquid Film in the Label text field.
- **3** In the **Variable name** text field, type dom_cNaC1.
- 4 Locate the Expression section. In the Expression text field, type cNaC1.
- 5 Locate the Table and Window Settings section. Click + Add Table.
- 6 Click + Add Plot Window.

Variables - Film, Variables - Global, Variables - Rates, Variables - Steel, Variables - Zn

- I In the Model Builder window, under Component I (compl) > Definitions, Ctrl-click to select Variables - Global, Variables - Rates, Variables - Film, Variables - Zn, and Variables -Steel.
- 2 Right-click and choose Group.

Group - Variables

In the Settings window for Group, type Group - Variables in the Label text field.

Interpolation - Relative Humidity (RH_ACT), Interpolation - Spray (spray_ACT), Interpolation - Temperature (T_ACT)

- I In the Model Builder window, under Component I (compl) > Definitions, Ctrl-click to select Interpolation - Relative Humidity (RH_ACT), Interpolation - Temperature (T_ACT), and Interpolation - Spray (spray_ACT).
- 2 Right-click and choose Group.

Group - Interpolations

In the Settings window for Group, type Group - Interpolations in the Label text field.

Domain Probe - Average NaCl Concentration in Liquid Film (dom_cNaCl), Domain Probe - Maximum Zn Coating Thickness Loss (dom_d_Zn), Domain Probe - Maximum pH (dom_maxpH), Domain Probe - Minimum pH (dom_minpH), Global Variable Probe -Total Oxygen Reduction Current (var_l_red), Global Variable Probe - Total Zinc Metal Dissolution Current (var_l_ox), Global Variable Probe - Zn Metal (var_m_Zn), Global Variable Probe - ZnO (var_ZnO)

In the Model Builder window, under Component I (comp1) > Definitions, Ctrl-click to select Global Variable Probe - ZnO (var_ZnO), Global Variable Probe The Model (var. 2n) Clobal Variable Probe -

Zn Metal (var_m_Zn), Global Variable Probe -

Total Zinc Metal Dissolution Current (var_l_ox), Global Variable Probe -Total Oxygen Reduction Current (var_l_red), Domain Probe -Maximum Zn Coating Thickness Loss (dom_d_Zn), Domain Probe -Maximum pH (dom_maxpH), Domain Probe - Minimum pH (dom_minpH), and Domain Probe - Average NaCl Concentration in Liquid Film (dom_cNaCl).

2 Right-click and choose Group.

Group - Probes

In the Settings window for Group, type Group - Probes in the Label text field.

Integration - Steel (intop_steel), Integration - Wetted Surface (intop_wet), Integration - Zn (intop_zinc)

- I In the Model Builder window, under Component I (compl) > Definitions, Ctrl-click to select Integration - Zn (intop_zinc), Integration - Steel (intop_steel), and Integration -Wetted Surface (intop_wet).
- 2 Right-click and choose Group.

Group - Integrations

In the Settings window for Group, type Group - Integrations in the Label text field.

Set the solver time-stepping to a value lower than the distinct periods of the ACT. In this case, 0.1 h is well below 3 h.

STUDY I

Step 1: Current Distribution Initialization

- I In the Model Builder window, under Study I click Step I: Current Distribution Initialization.
- 2 In the Settings window for Current Distribution Initialization, locate the Physics and Variables Selection section.
- 3 In the Solve for column of the table, under Component I (compl), clear the checkbox for Dissolving and Depositing Species (c).

Step 2: Time Dependent

- I In the Model Builder window, click Step 2: Time Dependent.
- 2 In the Settings window for Time Dependent, locate the Study Settings section.
- 3 From the Time unit list, choose h.
- 4 Click Range.
- 5 In the Range dialog, type 1.5 in the Step text field.
- 6 In the **Stop** text field, type 24*7.
- 7 Click Replace.

Solution I (soll)

- I In the Study toolbar, click **The Show Default Solver**.
- 2 In the Model Builder window, under Study I > Solver Configurations > Solution I (soll) click Time-Dependent Solver I.
- 3 In the Settings window for Time-Dependent Solver, locate the General section.
- **4** From the Times to store list, choose Steps taken by solver closest to output times.
- **5** Click to expand the **Time Stepping** section. From the **Maximum step constraint** list, choose **Constant**.
- 6 In the Home toolbar, click **=** Compute.

RESULTS

Start polishing the probe plots. All of these are found in the model documentation.

Mass Change on Full Sample

- I In the Model Builder window, under Results click Probe Plot Group I.
- 2 In the Settings window for ID Plot Group, type Mass Change on Full Sample in the Label text field.
- 3 Click to expand the Title section. From the Title type list, choose Manual.
- 4 Locate the Plot Settings section.
- 5 Select the y-axis label checkbox. In the associated text field, type Mass Change (mg).
- 6 Select the Two y-axes checkbox.
- 7 Select the Secondary y-axis label checkbox. In the associated text field, type Relative humidity (1).
- 8 Locate the Legend section. From the Layout list, choose Outside graph axis area.
- **9** From the **Position** list, choose **Top**.

Probe Table Graph 1

- I In the Model Builder window, expand the Mass Change on Full Sample node, then click Probe Table Graph I.
- 2 In the Settings window for Table Graph, click to expand the Legends section.
- 3 From the Legends list, choose Manual.
- 4 In the table, enter the following settings:

Legends

Total mass increase of ZnO

Total mass decrease of Zn metal

Mass Change on Full Sample

In the Mass Change on Full Sample toolbar, click () Global.

Global I

- I In the Settings window for Global, locate the Data section.
- 2 From the Dataset list, choose Study I/Solution I (soll).
- 3 From the Time selection list, choose Interpolated.
- 4 In the **Times (h)** text field, type range(0,0.1,24*7).
- 5 Locate the y-Axis section. Select the Plot on secondary y-axis checkbox.

6 Locate the y-Axis Data section. In the table, enter the following settings:

Expression	Unit	Description
RH_ACT(t)	1	Interpolation - Relative Humidity

- 7 Click to expand the Legends section. From the Legends list, choose Manual.
- 8 In the table, enter the following settings:

Legends

Relative humidity

9 In the Mass Change on Full Sample toolbar, click 💿 Plot.

Total Current on Full Sample

- I In the Model Builder window, under Results click Probe Plot Group 2.
- 2 In the Settings window for ID Plot Group, type Total Current on Full Sample in the Label text field.
- 3 Locate the Plot Settings section.
- 4 Select the **y-axis label** checkbox. In the associated text field, type Total current (\mu A).
- 5 Locate the Legend section. From the Layout list, choose Outside graph axis area.
- 6 From the **Position** list, choose **Top**.

Probe Table Graph 1

- I In the Model Builder window, expand the Total Current on Full Sample node, then click Probe Table Graph I.
- 2 In the Settings window for Table Graph, locate the Legends section.
- 3 From the Legends list, choose Manual.
- **4** In the table, enter the following settings:

Legends

Zn metal dissolution

Oxygen reduction on steel

5 In the Total Current on Full Sample toolbar, click 🗿 Plot.

Maximum Decrease Coating Thickness

I In the Model Builder window, under Results click Probe Plot Group 3.

- 2 In the Settings window for ID Plot Group, type Maximum Decrease Coating Thickness in the Label text field.
- 3 Locate the Plot Settings section.
- 4 Select the y-axis label checkbox. In the associated text field, type Thickness decrease (μm).
- 5 Locate the Legend section. Clear the Show legends checkbox.
- 6 In the Maximum Decrease Coating Thickness toolbar, click 💽 Plot.

pH Limits in Film

- I In the Model Builder window, under Results click Probe Plot Group 4.
- 2 In the Settings window for ID Plot Group, type pH Limits in Film in the Label text field.
- 3 Locate the Plot Settings section.
- 4 Select the y-axis label checkbox. In the associated text field, type pH (-).
- 5 Locate the Legend section. From the Layout list, choose Outside graph axis area.
- 6 From the **Position** list, choose **Top**.

Probe Table Graph 1

- I In the Model Builder window, expand the pH Limits in Film node, then click Probe Table Graph I.
- 2 In the Settings window for Table Graph, locate the Legends section.
- 3 From the Legends list, choose Manual.
- **4** In the table, enter the following settings:

Legends

Maximum

Minimum

5 In the pH Limits in Film toolbar, click 💿 Plot.

Average NaCl Concentration and Liquid Film Thickness

- I In the Model Builder window, under Results click Probe Plot Group 5.
- 2 In the **Settings** window for **ID Plot Group**, type Average NaCl Concentration and Liquid Film Thickness in the **Label** text field.
- **3** Locate the **Title** section. From the **Title type** list, choose **None**.
- 4 Locate the **Plot Settings** section.

- 5 Select the y-axis label checkbox. In the associated text field, type Average NaCl concentration (mol/m³).
- 6 Select the Two y-axes checkbox.
- 7 Select the Secondary y-axis label checkbox. In the associated text field, type Liquid film thickness (\mu m).
- 8 Locate the Grid section. Clear the Show grid checkbox.
- 9 Locate the Legend section. From the Layout list, choose Outside graph axis area.

IO From the **Position** list, choose **Top**.

Probe Table Graph 1

- In the Model Builder window, expand the Average NaCl Concentration and Liquid Film Thickness node, then click Probe Table Graph 1.
- 2 In the Settings window for Table Graph, locate the Legends section.
- 3 From the Legends list, choose Manual.
- **4** In the table, enter the following settings:

Legends

NaCl concentration

Probe Table Graph 1.1

- I Right-click Probe Table Graph I and choose Duplicate.
- 2 In the Settings window for Table Graph, click to expand the Preprocessing section.
- 3 Find the y-axis columns subsection. From the Range list, choose Manual.
- 4 In the y minimum text field, type cNaCl_deliq.
- **5** Locate the **Coloring and Style** section. Find the **Line style** subsection. From the **Line** list, choose **Dotted**.
- 6 From the Color list, choose White.
- 7 Locate the Legends section. Clear the Show legends checkbox.

Average NaCl Concentration and Liquid Film Thickness

In the Average NaCl Concentration and Liquid Film Thickness toolbar, click 🗁 Global.

Global I

- I In the Settings window for Global, locate the Data section.
- 2 From the Dataset list, choose Study I/Solution I (soll).
- 3 Locate the y-Axis section. Select the Plot on secondary y-axis checkbox.

4 Locate the y-Axis Data section. In the table, enter the following settings:

Expression	Unit	Description
d_film	um	Liquid film thickness with RH

5 Locate the Legends section. From the Legends list, choose Manual.

6 In the table, enter the following settings:

Legends

Film thickness

Global 2

- I Right-click Results > Average NaCl Concentration and Liquid Film Thickness > Global I and choose Duplicate.
- 2 In the Settings window for Global, click to expand the Coloring and Style section.
- 3 Find the Line style subsection. From the Line list, choose Dotted.
- 4 From the Color list, choose White.
- 5 Locate the Legends section. Clear the Show legends checkbox.

Filter I

- I In the Average NaCl Concentration and Liquid Film Thickness toolbar, click 🐺 Filter.
- 2 In the Settings window for Filter, locate the Point Selection section.
- 3 In the Logical expression for inclusion text field, type RH_ACT(t)<RH_deliq.
- 4 In the Average NaCl Concentration and Liquid Film Thickness toolbar, click 💽 Plot.

АСТ

- I In the **Results** toolbar, click \sim **ID Plot Group**.
- 2 In the Settings window for ID Plot Group, type ACT in the Label text field.
- **3** Locate the **Title** section. From the **Title type** list, choose **None**.
- 4 Locate the Data section. From the Time selection list, choose Interpolated.
- 5 In the Times (h) text field, type range(0,0.1,24*7).
- 6 Locate the Plot Settings section. Select the x-axis label checkbox.
- 7 Select the y-axis label checkbox. In the associated text field, type Relative humidity (1).
- 8 Select the Two y-axes checkbox.
- 9 Select the Secondary y-axis label checkbox. In the associated text field, type Temperature (K).

IO Locate the **Grid** section. Clear the **Show grid** checkbox.

II Locate the Legend section. From the Layout list, choose Outside graph axis area.

12 From the **Position** list, choose **Top**.

The following steps creates a plot illustrating the ACT that is simulated.

Global I

- I In the ACT toolbar, click (Global.
- 2 In the Settings window for Global, locate the y-Axis Data section.
- **3** In the table, enter the following settings:

Expression	Unit	Description
RH_ACT(t)	1	Interpolation - Relative Humidity

4 Locate the Legends section. From the Legends list, choose Manual.

5 In the table, enter the following settings:

Legends

Relative humidity

Global 2

- I Right-click Global I and choose Duplicate.
- 2 In the Settings window for Global, locate the Coloring and Style section.
- **3** Find the Line style subsection. From the Line list, choose None.
- 4 From the Color list, choose Black.
- 5 Find the Line markers subsection. From the Marker list, choose Point.
- 6 Locate the Legends section. Clear the Show legends checkbox.

Filter I

- I In the **ACT** toolbar, click **\ Filter**.
- 2 In the Settings window for Filter, locate the Point Selection section.
- 3 In the Logical expression for inclusion text field, type spray_ACT(t)>0.5.

Global I, Global 2

- I In the Model Builder window, under Results > ACT, Ctrl-click to select Global I and Global 2.
- 2 Right-click and choose **Duplicate**.

Global 3

I In the Settings window for Global, locate the y-Axis section.

2 Select the Plot on secondary y-axis checkbox.

3 Locate the y-Axis Data section. In the table, enter the following settings:

Expression	Unit	Description
T_ACT(t)	К	Interpolation - Temperature

4 Locate the Legends section. In the table, enter the following settings:

Legends

Temperature

Global 4

I In the Model Builder window, click Global 4.

2 In the Settings window for Global, locate the y-Axis section.

3 Select the **Plot on secondary y-axis** checkbox.

4 Locate the y-Axis Data section. In the table, enter the following settings:

Expression	Unit	Description
T_ACT(t)	к	Interpolation - Temperature

5 In the ACT toolbar, click **O** Plot.

The following steps create 2D plots for better display of local variations at the surface. Use a **Sector 2D** dataset that can visualize the results over the whole sample surface.

Sector 2D I

- I In the **Results** toolbar, click **More Datasets** and choose **Sector 2D**.
- 2 In the Settings window for Sector 2D, locate the Axis Data section.
- **3** In the **X** text field, type w_sample/2.
- **4** In the **Y** text field, type h_sample/2.
- 5 Locate the Symmetry section. In the Number of sectors text field, type 4.
- 6 From the Transformation list, choose Rotation and reflection.

pH Full Sample

- I In the **Results** toolbar, click **2D Plot Group**.
- 2 In the Settings window for 2D Plot Group, type pH Full Sample in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose None.

- 4 Click to expand the Title section. From the Title type list, choose Manual.
- 5 In the Title text area, type pH on sample at different times.
- 6 Clear the **Parameter indicator** text field.
- 7 Locate the Color Legend section. From the Position list, choose Bottom.
- 8 Click to expand the Plot Array section. Select the Enable checkbox.
- 9 From the Array shape list, choose Square.

Surface 1

- I In the pH Full Sample toolbar, click **A** Surface with Height.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Dataset list, choose Sector 2D I.
- 4 From the Time (h) list, choose 1.5543.
- 5 Click to expand the Range section. Select the Manual color range checkbox.
- 6 In the Minimum text field, type 6.7.
- 7 In the Maximum text field, type 10.7.
- 8 Locate the Coloring and Style section. From the Color table list, choose PrismDark.
- 9 From the Color table transformation list, choose Reverse.

Height Expression 1

- I In the Model Builder window, expand the Surface I node, then click Height Expression I.
- 2 In the Settings window for Height Expression, locate the Axis section.
- 3 Clear the Show height axis checkbox.

Surface 2

- I In the Model Builder window, under Results > pH Full Sample right-click Surface I and choose Duplicate.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Time (h) list, choose 10.572.
- 4 Click to expand the Inherit Style section. From the Plot list, choose Surface I.

Surface 3

- I Right-click Surface 2 and choose Duplicate.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Time (h) list, choose 18.023.

Surface 4

- I Right-click Surface 3 and choose Duplicate.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Time (h) list, choose 21.018.

Surface 5

- I Right-click Surface 4 and choose Duplicate.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Time (h) list, choose 52.522.

Surface 6

- I Right-click Surface 5 and choose Duplicate.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Time (h) list, choose Last (168).
- **4** Click the **Zoom Extents** button in the **Graphics** toolbar.

Annotation I

- I In the Model Builder window, right-click pH Full Sample and choose Annotation.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Dataset list, choose Sector 2D I.
- 4 From the Time (h) list, choose 1.5543.
- **5** Locate the **Annotation** section. In the **Text** text field, type eval(t,h) h.
- 6 Click to expand the Advanced section. Clear the Show trailing zeros checkbox.
- 7 In the **Precision** text field, type 2.
- 8 Locate the Coloring and Style section. Clear the Show point checkbox.
- 9 From the Anchor point list, choose Upper right.

10 Click to expand the Plot Array section. Select the Manual indexing checkbox.

- II In the **Row index** text field, type 1.
- **12** In the **Column index** text field, type 1.

Annotation 2

- I Right-click Annotation I and choose Duplicate.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Time (h) list, choose 10.572.
- 4 Locate the Advanced section. In the Precision text field, type 3.

5 Locate the Plot Array section. In the Column index text field, type 2.

Annotation 3

- I Right-click Annotation 2 and choose Duplicate.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Time (h) list, choose 18.023.
- 4 Locate the Plot Array section. In the Column index text field, type 3.

Annotation 4

- I Right-click Annotation 3 and choose Duplicate.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Time (h) list, choose 21.018.
- 4 Locate the Plot Array section. In the Row index text field, type 2.
- **5** In the **Column index** text field, type **1**.

Annotation 5

- I Right-click Annotation 4 and choose Duplicate.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Time (h) list, choose 52.522.
- 4 Locate the Plot Array section. In the Column index text field, type 2.

Annotation 6

- I Right-click Annotation 5 and choose Duplicate.
- 2 In the Settings window for Annotation, locate the Data section.
- 3 From the Time (h) list, choose Last (168).
- 4 Locate the Plot Array section. In the Column index text field, type 3.
- 5 In the pH Full Sample toolbar, click 💽 Plot.
- 6 Click the **Zoom Extents** button in the **Graphics** toolbar.

Coating Thickness Decrease

- I In the **Results** toolbar, click **2D Plot Group**.
- 2 In the Settings window for 2D Plot Group, type Coating Thickness Decrease in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Sector 2D I.
- **4** Locate the **Title** section. From the **Title type** list, choose **Manual**.
- 5 In the Title text area, type Thickness decrease (\mu m).

6 Locate the Plot Settings section. Clear the Plot dataset edges checkbox.

Surface 1

- I In the Coating Thickness Decrease toolbar, click **M** Surface with Height.
- 2 In the Settings window for Surface, locate the Expression section.
- **3** In the **Expression** text field, type -d_Zn.
- 4 From the **Unit** list, choose µm.
- 5 Locate the Coloring and Style section. From the Color table list, choose Prionace.
- 6 From the Color table transformation list, choose Reverse.
- 7 In the Coating Thickness Decrease toolbar, click **Plot**.
- 8 Click the **Zoom Extents** button in the **Graphics** toolbar.

Corrosion Product Coverage Degree

- I In the Home toolbar, click 🚛 Add Plot Group and choose 2D Plot Group.
- 2 In the Settings window for 2D Plot Group, type Corrosion Product Coverage Degree in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Sector 2D I.
- 4 Locate the Title section. From the Title type list, choose Manual.
- 5 In the **Title** text area, type Coverage degree (1).
- 6 Locate the Plot Settings section. Clear the Plot dataset edges checkbox.

Surface 1

- I In the Corrosion Product Coverage Degree toolbar, click 💧 Surface with Height.
- 2 In the Settings window for Surface, locate the Expression section.
- **3** In the **Expression** text field, type theta.
- 4 Locate the Coloring and Style section. From the Color table list, choose Passiflora.
- 5 From the Color table transformation list, choose Reverse.

Height Expression 1

- I In the Model Builder window, expand the Surface I node, then click Height Expression I.
- 2 In the Settings window for Height Expression, locate the Axis section.
- 3 Clear the Show height axis checkbox.
- 4 In the Corrosion Product Coverage Degree toolbar, click 💿 Plot.
- **5** Click the **Comextents** button in the **Graphics** toolbar.

Finally, tidy some of the default plots or remove the ones considered redundant.

Dissolving and Depositing Species, Electrolyte Conductivity (el), Electrolyte Potential (el), Molar Concentration - CO3 (el), Molar Concentration - Cl (el), Molar Concentration - Na (el), Molar Concentration - Zn (el), pH (el)

- In the Model Builder window, under Results, Ctrl-click to select pH (el),
 Electrolyte Conductivity (el), Electrolyte Potential (el), Molar Concentration Zn (el),
 Molar Concentration CO3 (el), Molar Concentration Na (el), Molar Concentration Cl (el), and Dissolving and Depositing Species.
- 2 Right-click and choose Delete.