

Two-dimensional Fluid Simulation of an RF Capacitively Coupled Ar/H₂ Discharge

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Effect of small amount of H₂ added to Ar

Charge accumulation of the focus ring

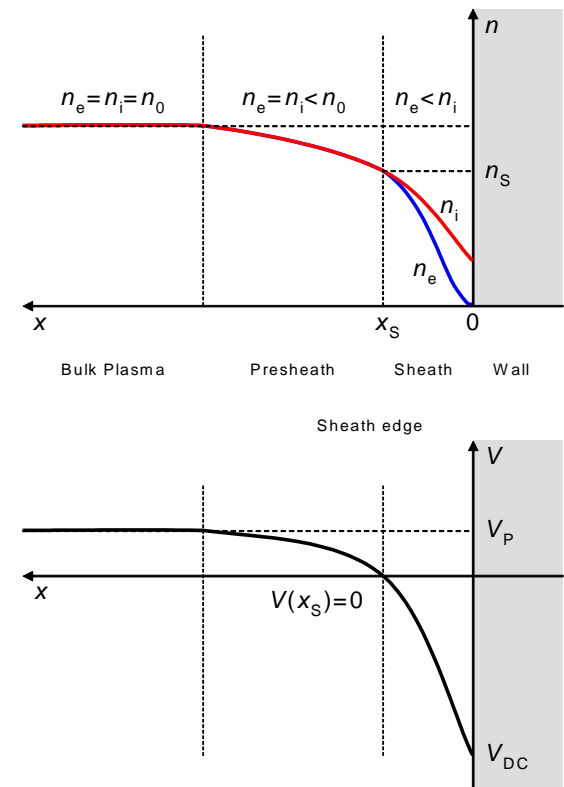
DC-bias generated by the blocking capacitor

- RF discharge and CCP plasma
- The computational model
- Results
 - CCP discharge structure
 - Effect of the focus ring and blocking capacitor
 - Comparison with the discharge of pure argon
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RF discharge and CCP plasma (1)

- RF discharge is important in the plasma CVD for the fabrication of thin film or in the field of plasma chemistry. In microwave plasma, a high ionization rate exists so that the plasma density is high.
- In RF plasma processing, the discharge threshold voltage is low. The discharge can be easily sustained and the electrode can be covered with dielectric materials.
- Types of RF plasma reactors
 - Inductively coupled plasma (ICP)
 - Microwave plasma (MWP)
 - Capacitively coupled plasma (CCP)
 - Combined ICP/CCP reactor

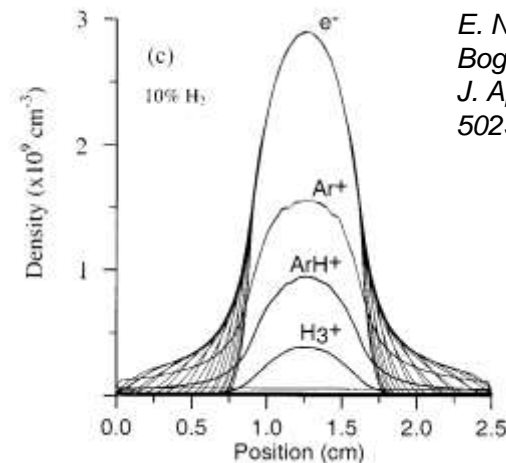
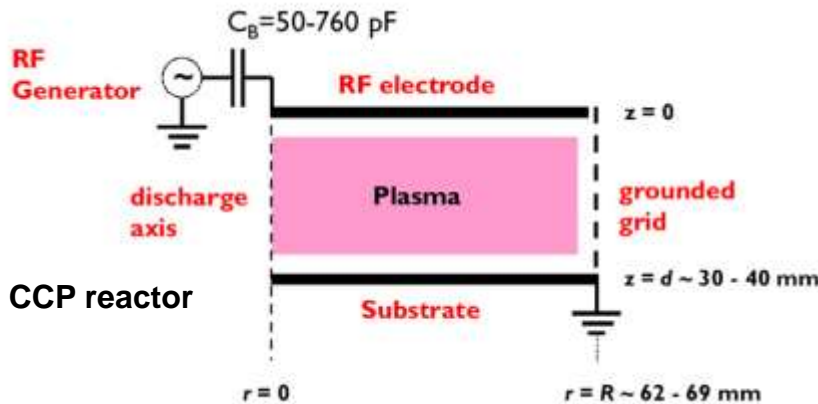
Interaction between plasma and wall



RF discharge and CCP plasma (2)

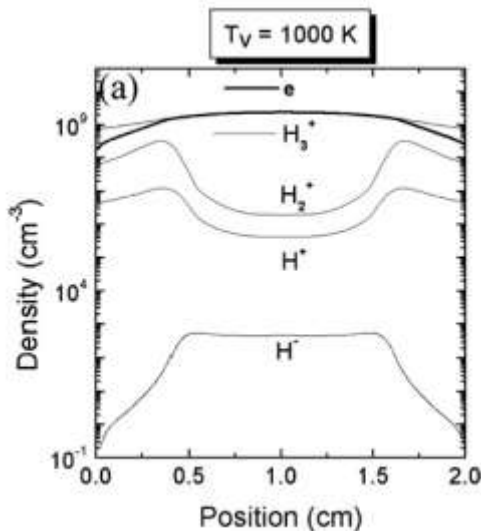
Capacitively coupled radio-frequency discharges are still among the most powerful and flexible plasma reactors, widely used both in research and in industry.

One-dimensional model

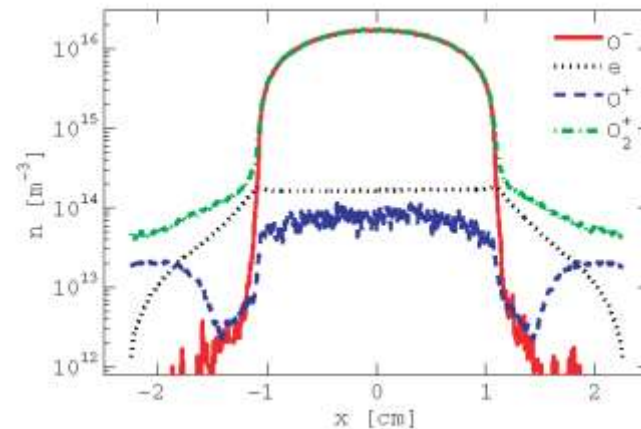


E. Neyts, M. Yan, A. Bogaerts, and R. Gijbels: *J. Appl. Phys.*, **93** (2003) 5025-5033

L. L. Alves and L. Marques: *Plasma Phys. Control. Fusion* **54** (2012) 124012 (8pp)



B. Kalache, T. Novikova, et al.: *J. Phys. D: Appl. Phys.* **37** (2004) 1765-1773

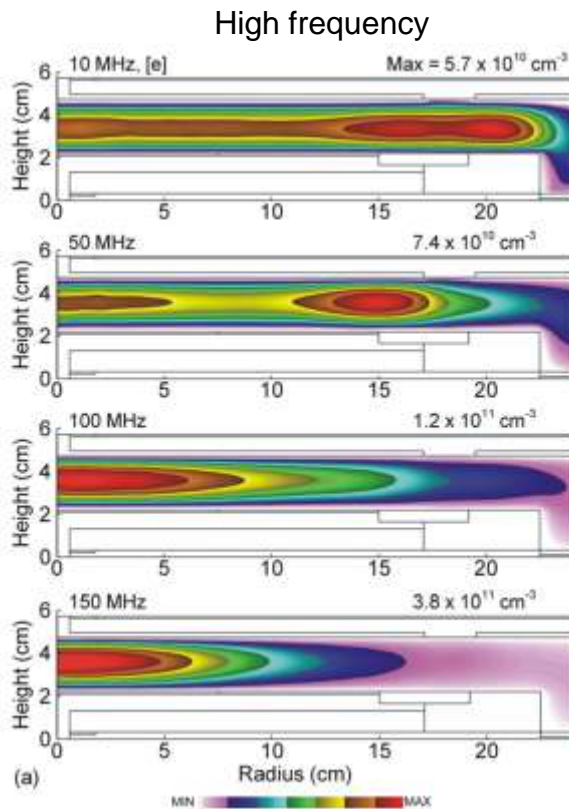


J. T. Gudmundsson, E. Kawamura and M. A. Lieberman: *Plasma Sources Sci. Technol.* **22** (2013) 035011 (11pp)

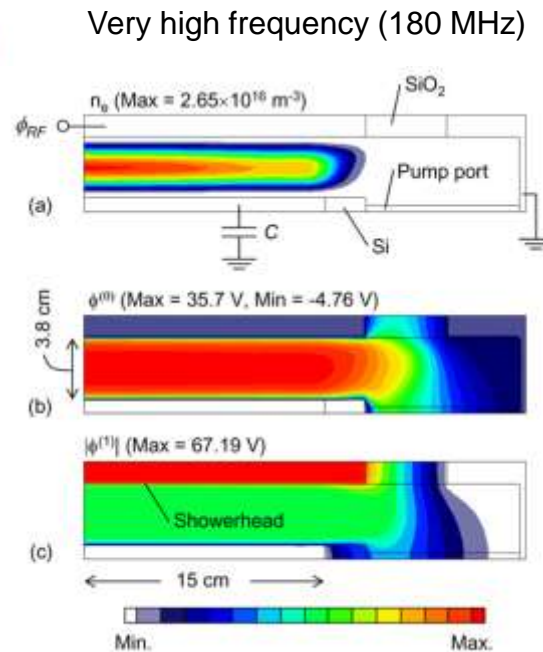
RF discharge and CCP plasma (3)

Among the different modelling approaches available to characterize CCP discharges, two-dimensional fluid models provide a good compromise solution within acceptable calculation runtimes.

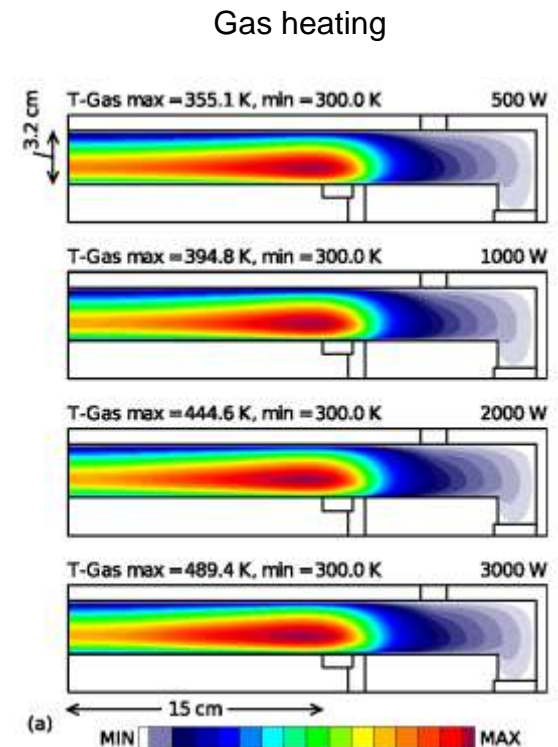
Two-dimensional model



Y. Yang and M. J. Kushner: *Plasma Sources Sci. Technol.* **19** (2010) 055011 (17pp)



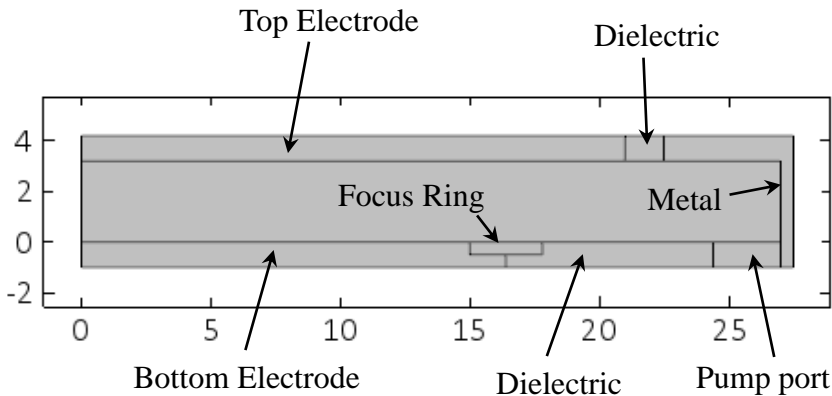
S. Rauf, K. Bera and K. Collins: *Plasma Sources Sci. Technol.* **17** (2008) 035003 (9pp)



A. Agarwal, S. Rauf and K. Collins: *Plasma Sources Sci. Technol.* **21** (2012) 055012 (12pp)

The computational model (1)

Model geometry



Computational conditions:

- Gases: Ar/H₂ mixtures (pure Ar, 1%H₂)
- Species: e⁻, Ar, H₂, Ar⁺, H⁺, H₂⁺, H₃⁺, ArH⁺, Ar^{*}, H, H(2p), H(2s)
- RF frequency: 13.56 MHz
- RF voltage: 200 V, applied to the bottom electrode
- Temperature: 300 K
- Gas pressure: 100 Pa
- Inter-electrode gap: 3.2 cm
- Blocking capacitor: 100 nF
- Focus ring: Silicon
- Dielectric: SiO₂

No.	Reaction
1	Ar + e ⁻ → Ar + e ⁻
2	Ar + e ⁻ → Ar [*] + e ⁻
3	Ar + e ⁻ → Ar ⁺ + 2e ⁻
4	Ar [*] + e ⁻ → Ar ⁺ + 2e ⁻
5	Ar [*] + Ar [*] → Ar ⁺ + Ar + e ⁻
6	Ar [*] + Ar → Ar + Ar
7	H ₂ + e ⁻ → H ₂ + e ⁻
8	H ₂ + e ⁻ → H + H + e ⁻
9	H ₂ + e ⁻ → H + H(2s) + e ⁻
10	H ₂ + e ⁻ → H(2p) + H(2s) + e ⁻
11	H ₂ + e ⁻ → H ₂ ⁺ + 2e ⁻
12	H ₂ + e ⁻ → H + H ⁺ + 2e ⁻
13	H ₂ ⁺ + e ⁻ → H ⁺ + H + e ⁻
14	H ₃ ⁺ + e ⁻ → H ₂ + H
15	H ₂ + H ₂ ⁺ → H ₃ ⁺ + H
16	H + e ⁻ → H + e ⁻
17	H + e ⁻ → H(2p) + e ⁻
18	H + e ⁻ → H(2s) + e ⁻
19	H(2s) + e ⁻ → H(2p) + e ⁻
20	H + e ⁻ → H ⁺ + 2e ⁻
21	H(2s) + e ⁻ → H ⁺ + 2e ⁻
22	H(2p) → H + hν
23	Ar [*] + H ₂ → Ar + H + H
24	Ar ⁺ + H ₂ → Ar + H ₂ ⁺
25	Ar ⁺ + H ₂ → H + ArH ⁺
26	ArH ⁺ + H ₂ → H ₃ ⁺ + Ar
27	ArH ⁺ + e ⁻ → Ar + H
28	Ar [*] → Ar (wall loss)
29	H ₂ ⁺ → H ₂ (wall loss)
30	H ₃ ⁺ → H + H ₂ (wall loss)
31	H ⁺ → H (wall loss)
32	H 1/2H ₂ (wall loss)
33	H(2p) → H (wall loss)
34	H(2s) → H (wall loss)

The computational model (2)

Drift-diffusion equations for electrons

$$\frac{\partial}{\partial t}(n_e) + \nabla \cdot \Gamma_e = R_e$$

$$\Gamma_e = -n_e(\mu_e \mathbf{E}) - D_e \nabla n_e$$

Source term

$$R_e = \sum_{j=1}^M x_j k_j N_n n_e$$

Reaction rate $k_j = \gamma \int_0^{\infty} \epsilon \sigma_j(\epsilon) f(\epsilon) d\epsilon$

↑
Cross sections for electron collisions

$$\frac{\partial}{\partial t}(n_\epsilon) + \nabla \cdot \Gamma_\epsilon + \mathbf{E} \cdot \Gamma_\epsilon = R_\epsilon$$

$$\Gamma_\epsilon = -n_\epsilon(\mu_\epsilon \mathbf{E}) - D_\epsilon \nabla n_\epsilon$$

Source term

$$R_\epsilon = \sum_{j=1}^P x_j k_j N_n n_e \Delta \epsilon_j$$

$$\gamma = (2q/m)^{1/2}$$

Boundary conditions:

$$-\mathbf{n} \cdot \Gamma_e = \left(\frac{1}{2} v_{e,th} n_e \right) - \sum_p \gamma_p (\Gamma_p \cdot \mathbf{n})$$

$$-\mathbf{n} \cdot \Gamma_\epsilon = \left(\frac{5}{6} v_{e,th} n_\epsilon \right) - \sum_p \epsilon_p \gamma_p (\Gamma_p \cdot \mathbf{n})$$

The computational model (3)

Modified Maxwell-Stefan equation for ion and neutral species

$$\rho \frac{\partial}{\partial t} (w_k) + \rho (\mathbf{u} \cdot \nabla) w_k = \nabla \cdot \mathbf{j}_k + R_k$$

ここに

$$\mathbf{j}_k = \rho \omega_k \mathbf{v}_k \quad \mathbf{v}_k = \sum_{j=1}^Q \tilde{D}_{kj} \mathbf{d}_k - \frac{D_k^T}{\rho \omega_k} \nabla \ln T$$

$$\mathbf{d}_k = \frac{1}{cRT} \left[\nabla p_k - \omega_k \nabla p - \rho_k \mathbf{g}_k + \omega_k \sum_{j=1}^Q \rho_j \mathbf{g}_j \right]$$

Boundary conditions: $-\mathbf{n} \cdot \mathbf{j}_k = M_\omega R_k + M_\omega c_k Z \mu_k (\mathbf{E} \cdot \mathbf{n}) [Z_k \mu_k (\mathbf{E} \cdot \mathbf{n}) > 0]$

Poisson's equation

$$-\nabla \cdot \varepsilon_0 \varepsilon_r \nabla V = \rho \quad \rho = q \left(\sum_{k=1}^N Z_k n_k - n_e \right)$$

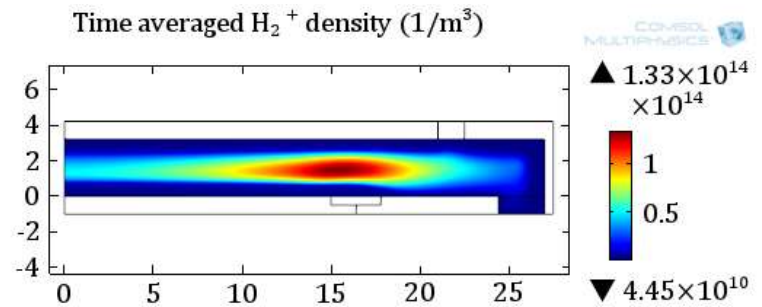
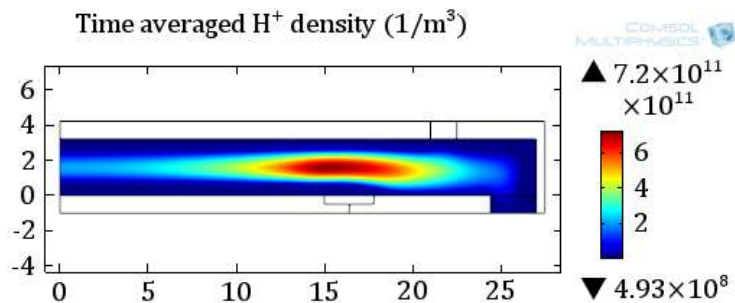
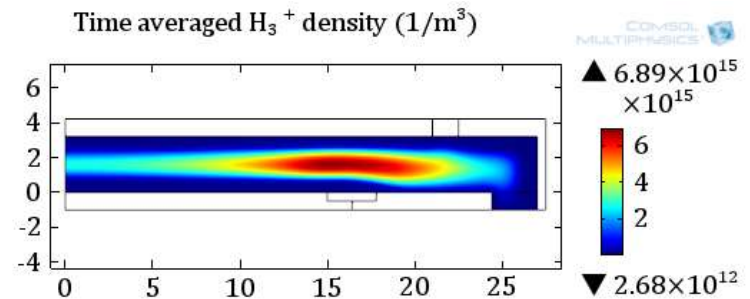
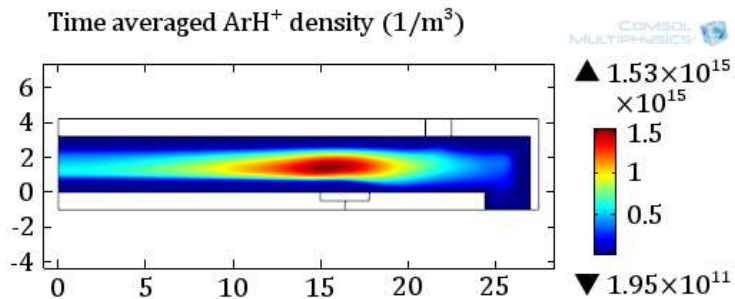
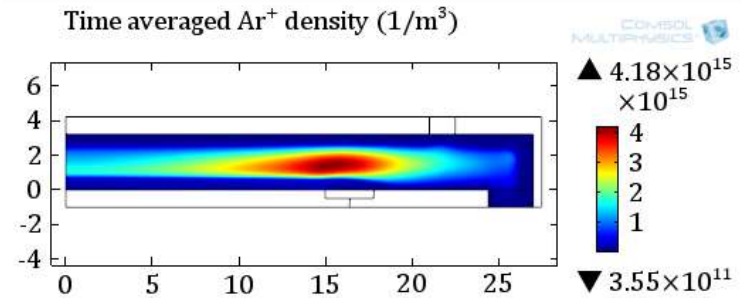
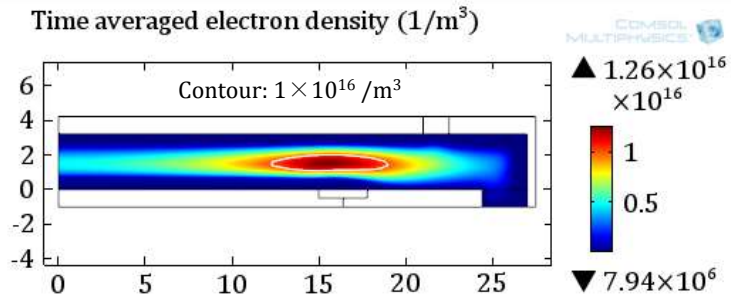
Charge accumulation on the dielectric surface :

$$\mathbf{n} \cdot (\mathbf{D}_1 - \mathbf{D}_2) = \rho_s \quad \frac{d\rho_s}{dt} = \mathbf{n} \cdot \mathbf{J}_i + \mathbf{n} \cdot \mathbf{J}_e$$

Results (1)

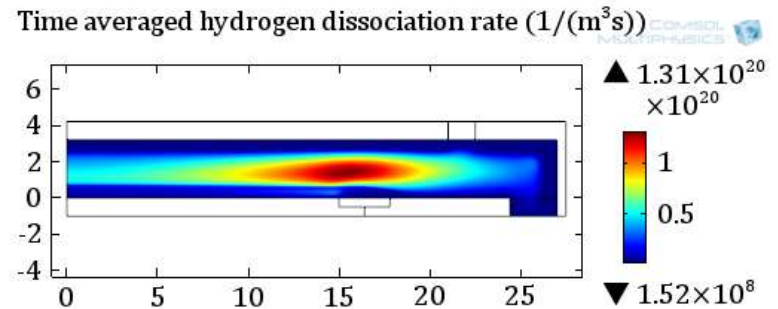
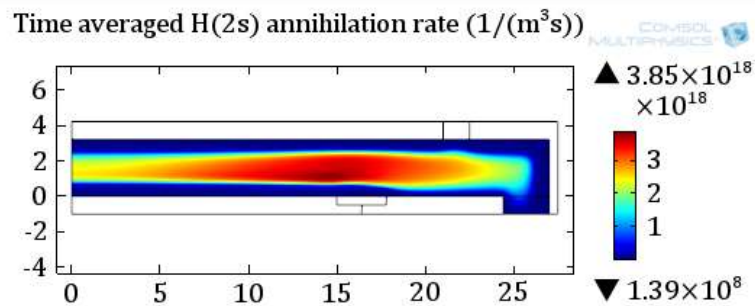
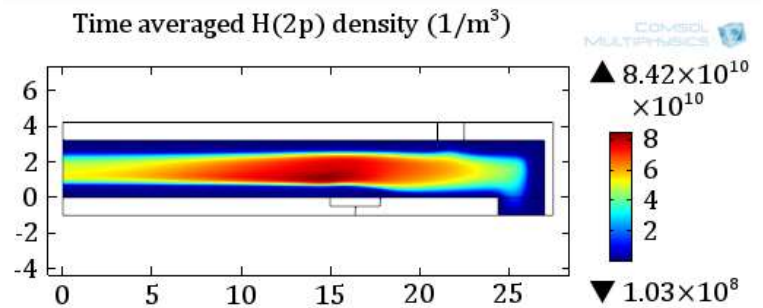
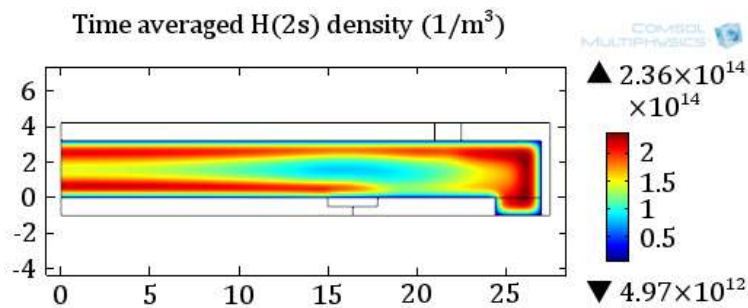
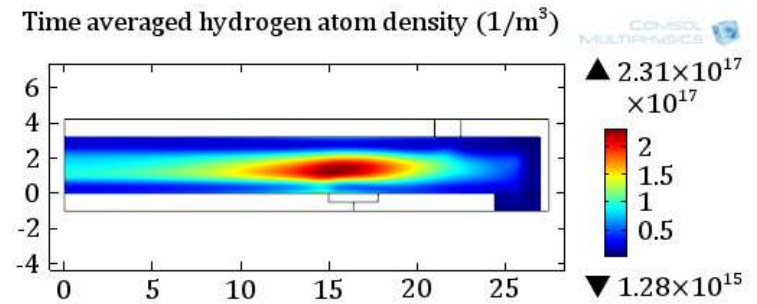
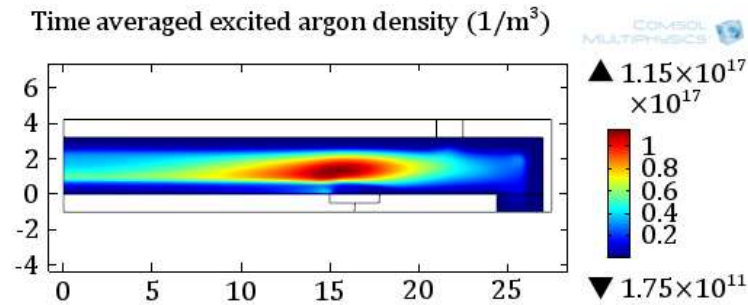
CCP discharge structure in Ar/1%H₂ mixture

Electron and ion densities



Results (2) CCP discharge structure in Ar/1%H₂ mixture

Neutral species densities

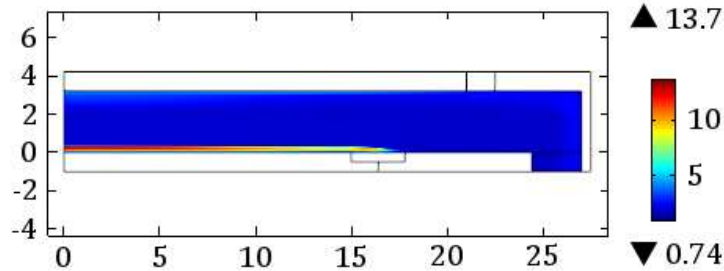


Results (3)

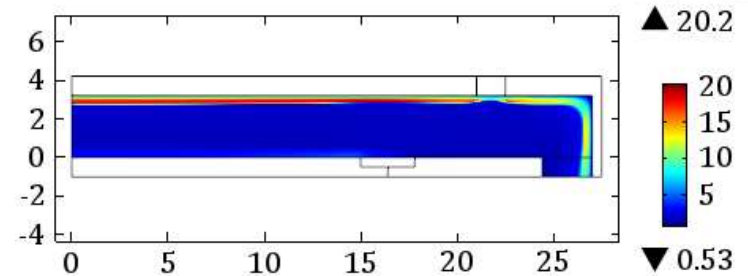
CCP discharge structure in Ar/1%H₂ mixture

Electron temperature

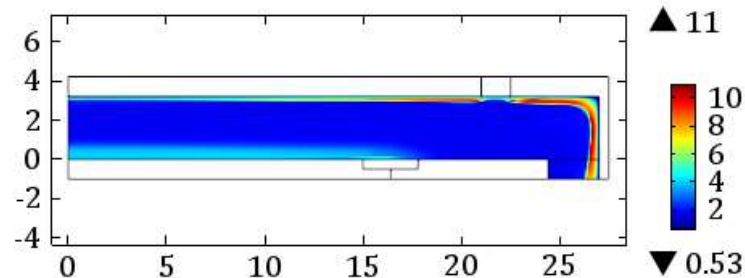
t = 0 Surface: Electron temperature (V)



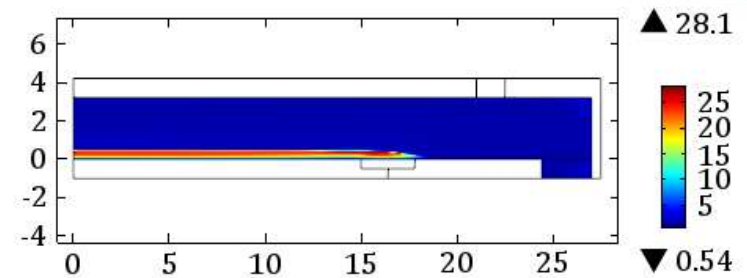
t = 1/4 T Surface: Electron temperature (V)



t = 1/2 T Surface: Electron temperature (V)



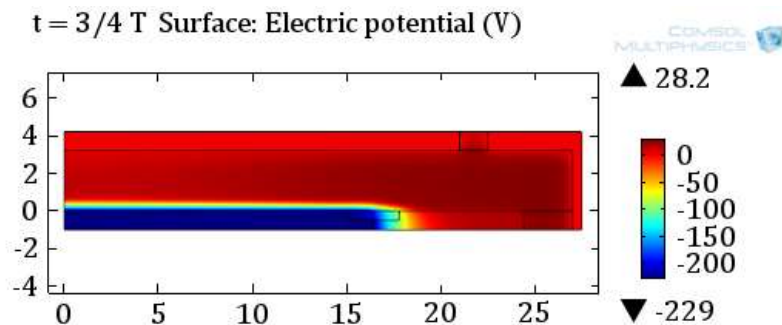
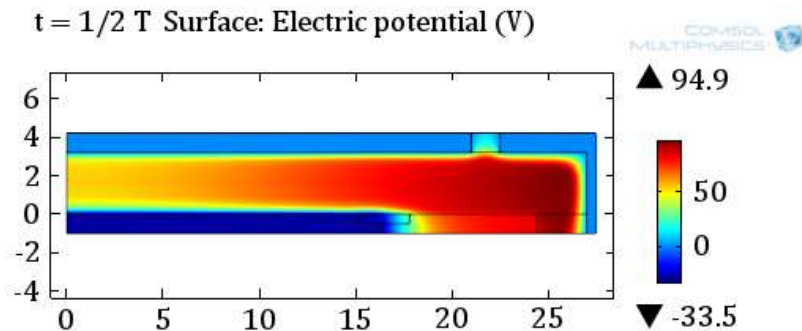
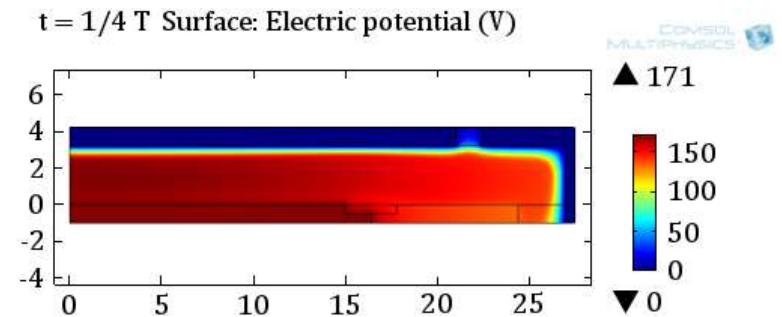
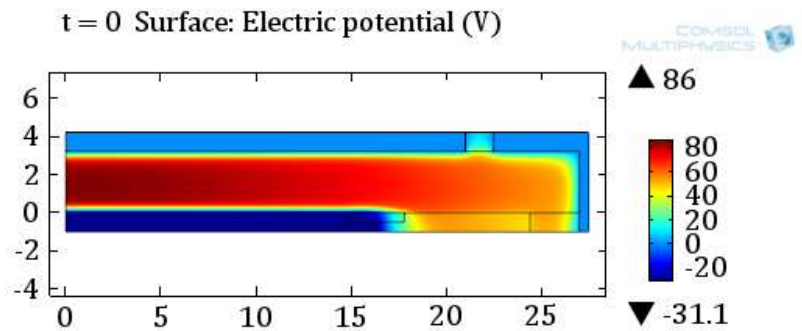
t = 3/4 T Surface: Electron temperature (V)



Results (4)

CCP discharge structure in Ar/1%H₂ mixture

Electric potential

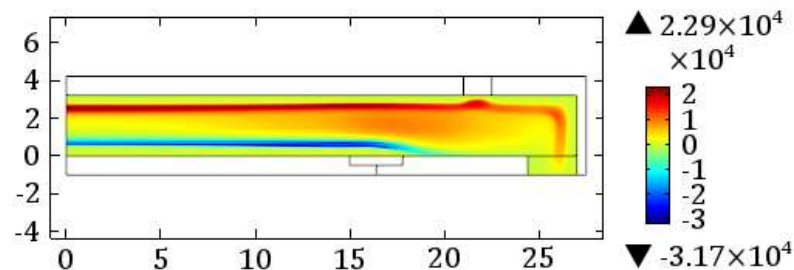


Results (5)

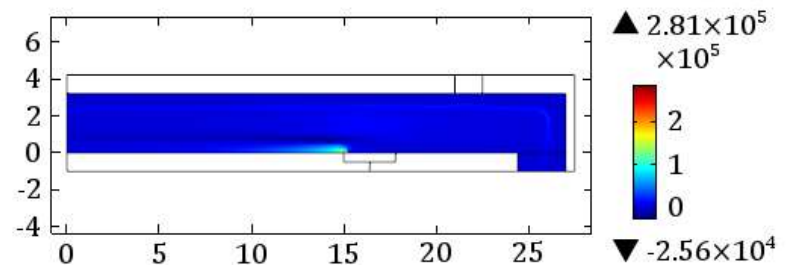
CCP discharge structure in Ar/1%H₂ mixture

Power deposition

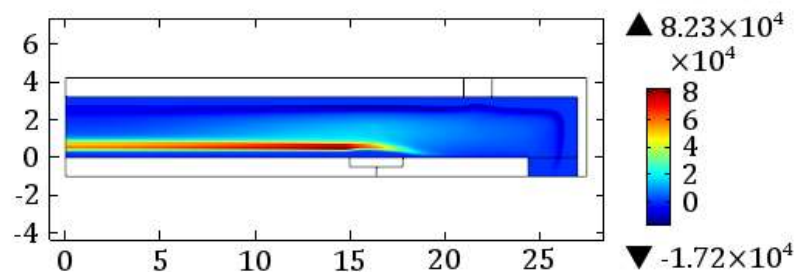
t = 0 Surface: Capacitive power deposition (W/m³)



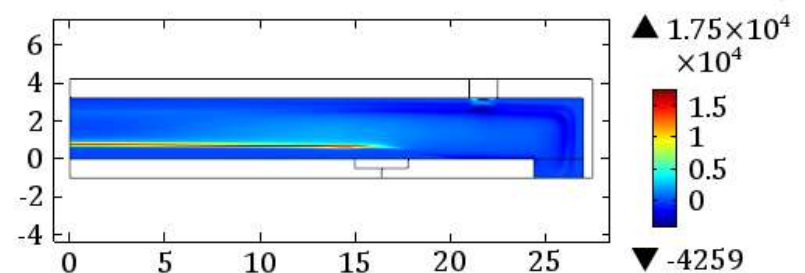
t = 1/4 T Surface: Capacitive power deposition (W/m³)



t = 1/2 T Surface: Capacitive power deposition (W/m³)



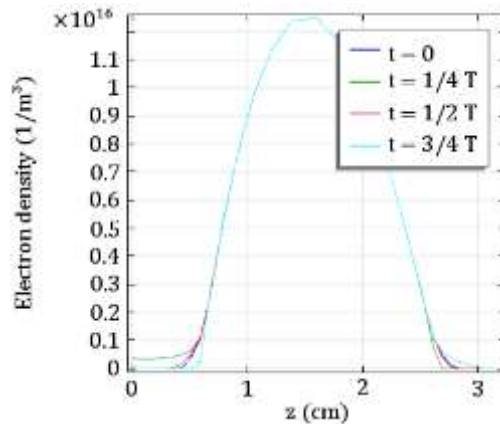
t = 3/4 T Surface: Capacitive power deposition (W/m³)



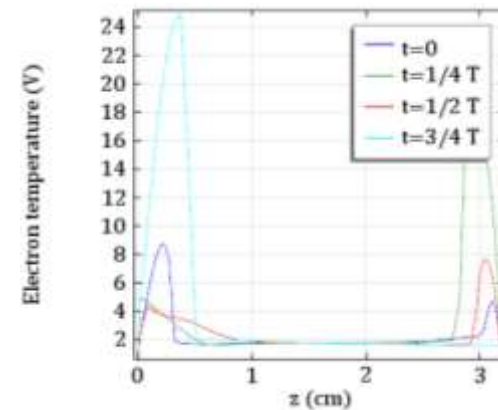
Results (6)

Discharge structure around the focus ring

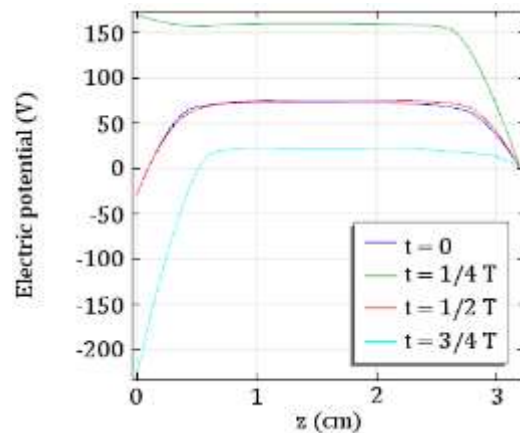
Electron density at $r = 15$ cm over the focus ring



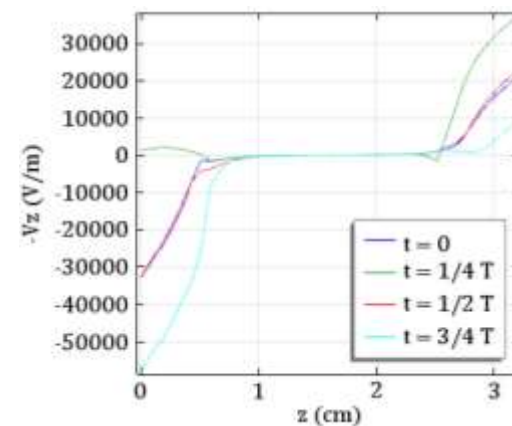
Electron temperature at $r = 15$ cm over the focus ring



Electric potential at $r = 15$ cm over the focus ring



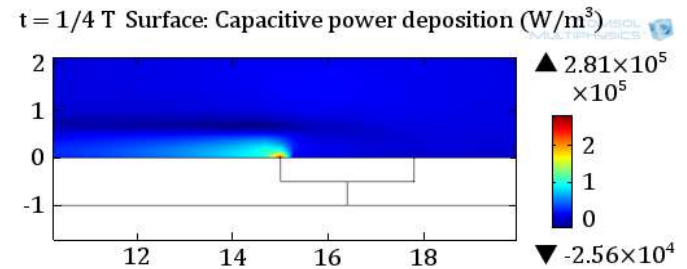
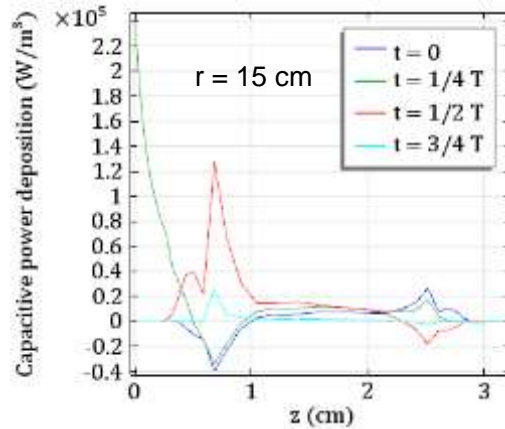
Electric field at $r = 15$ cm over the focus ring



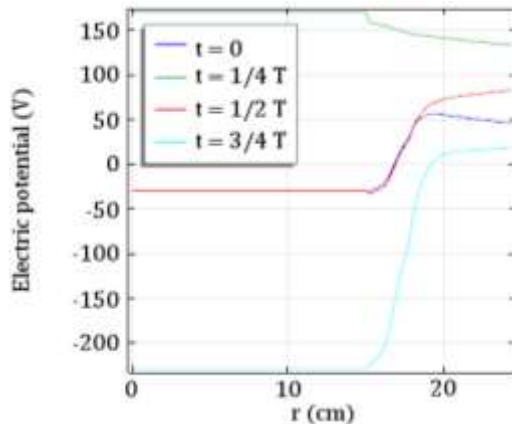
Results (7)

Effect of the focus ring and blocking capacitor

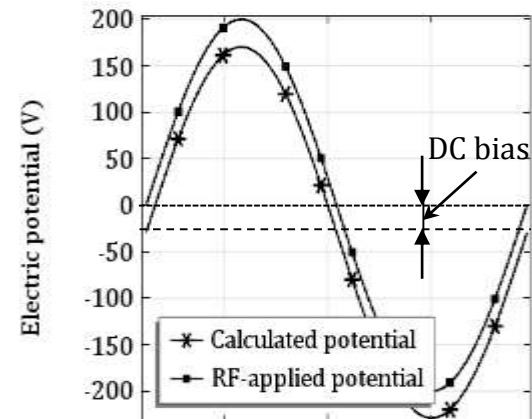
Power deposition around the focus ring



Electric potential along the surface of substrate and adjacent dielectrics

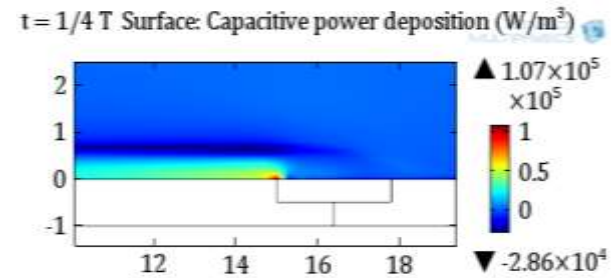
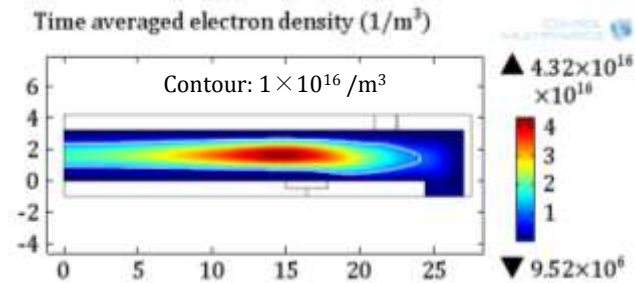


DC-bias generated by the blocking capacitor

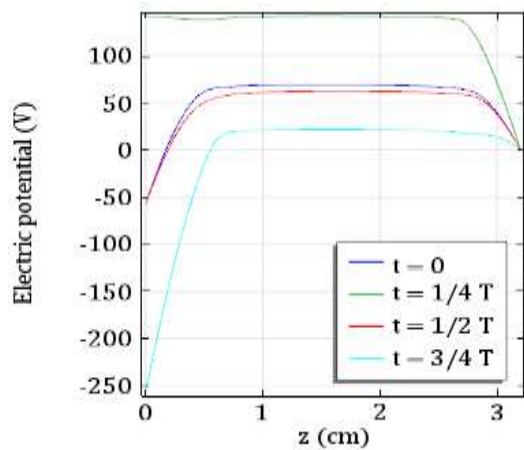


Results (8)

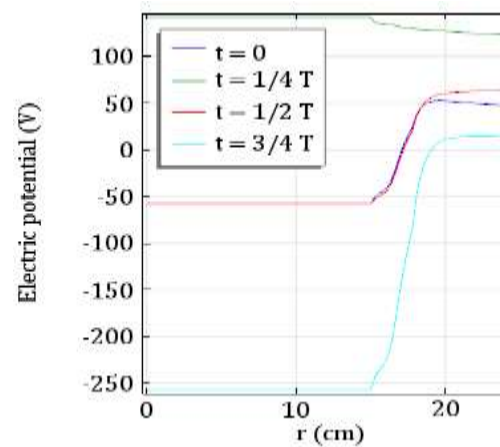
CCP discharge structure in pure Ar



Electric potential at $r = 15$ cm over the focus ring



Electric potential along the surface of substrate and adjacent dielectrics



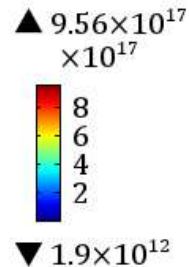
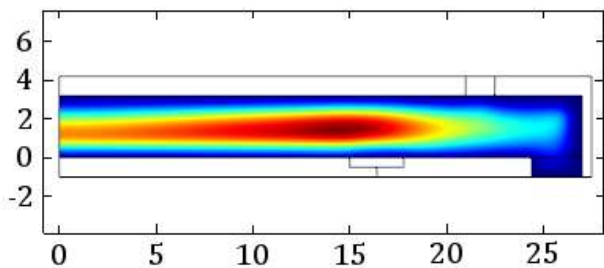
Results (9)

Comparison with the discharge in Ar/1%H₂ and pure Ar

Pure Ar

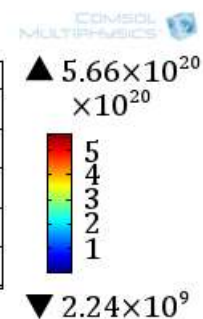
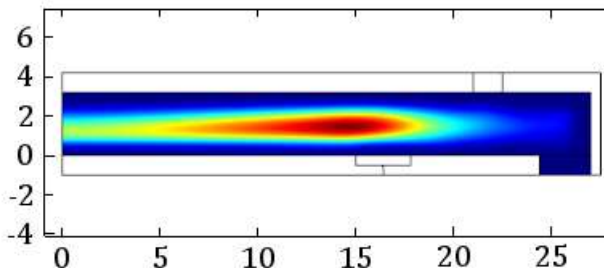
Excited argon density

Time averaged excited argon density (1/m³)



Pooling ionization rate

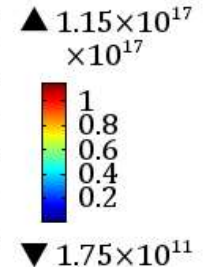
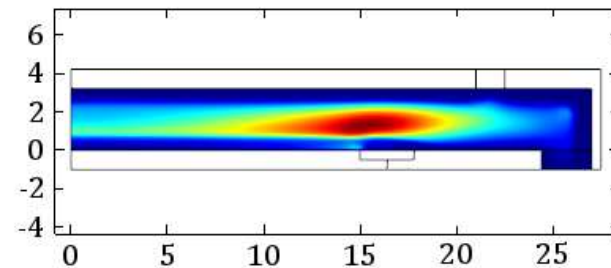
Time averaged ionization rate (1/(m³s))



Ar/1%H₂

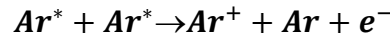
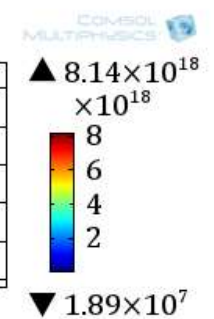
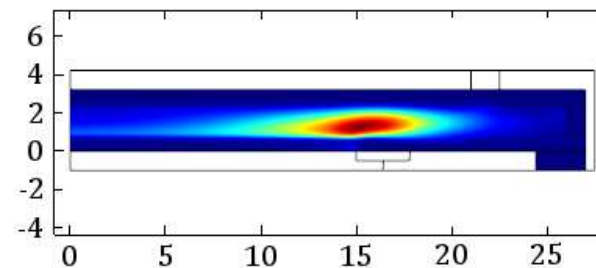
Excited argon density

Time averaged excited argon density (1/m³)



Pooling ionization rate

Time averaged ionization rate (1/(m³s))



Conclusions

- This paper presents the simulation results of low-pressure capacitively coupled RF plasmas in Ar/H₂.
- The addition of small amount of H₂ to Ar causes the electron density markedly decrease. The high electron density region is formed above the focus ring. The effect of the self DC-bias of the blocking capacitor is presented.
- It is found that with the increase of the amount of H₂ added to Ar, the density of metastable argon atoms is dramatically decreased. The pooling ionization rate due to the collisions among these atoms reduces down to 1.5% of that of pure argon.
- It could be concluded that the control of gas composition, focus ring and blocking capacitor would be very beneficial in finding the design parameters of RF CCP plasma reactors.

Thank you for your attention !



Questions & Comments ?