
European COMSOL Conference, Hannover, Germany 04.-06.11.2008

Modeling Polybenzimidazole/Phosphoric Acid Membrane Behaviour in a HTPEM Fuel Cell



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Introduction:

- University of Duisburg-Essen / ZBT-Duisburg GmbH
- High temperature PEM fuel cell (HTPEM)
- Review: 2D HTPEM model (European COMSOL Conference 2007)
- Objectives of this 3D study

Modeling efforts:

- Computational subdomains
- Governing equations
- Boundary conditions

Computational methodology:

- Meshing
- Solver settings and solution procedures

Results:

- Modeling base case operating conditions
- Experimental investigations
- Conclusion & outlook

University Duisburg-Essen: Hydrogen and fuel cell R&D since 1996

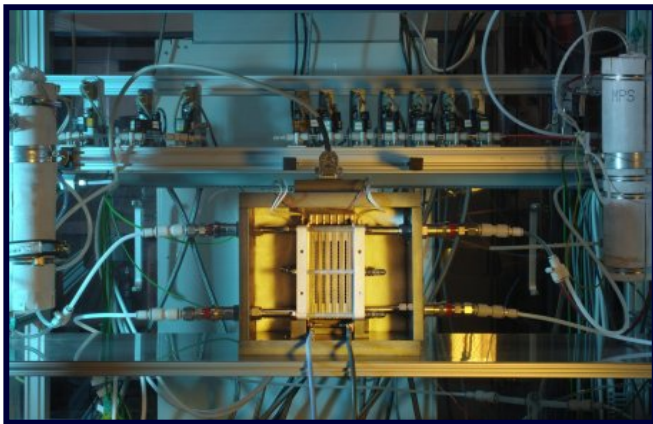


ZBT-Duisburg GmbH established in 2001

TAZ established in 2008



Hydrogen and fuel cell related activities in 7 divisions



Focusing the HTPEM technology:

- Bipolar-plate development
- Cell and stack design and operation
- System integration
- Theoretical analyses (e.g., modeling and simulation)

Fuel Cell - Device which electrochemically converts energy stored in a fuel and oxidant into electricity (e.g. oxygen/hydrogen half - cell reactions)

Some benefits of using higher operating temperatures:

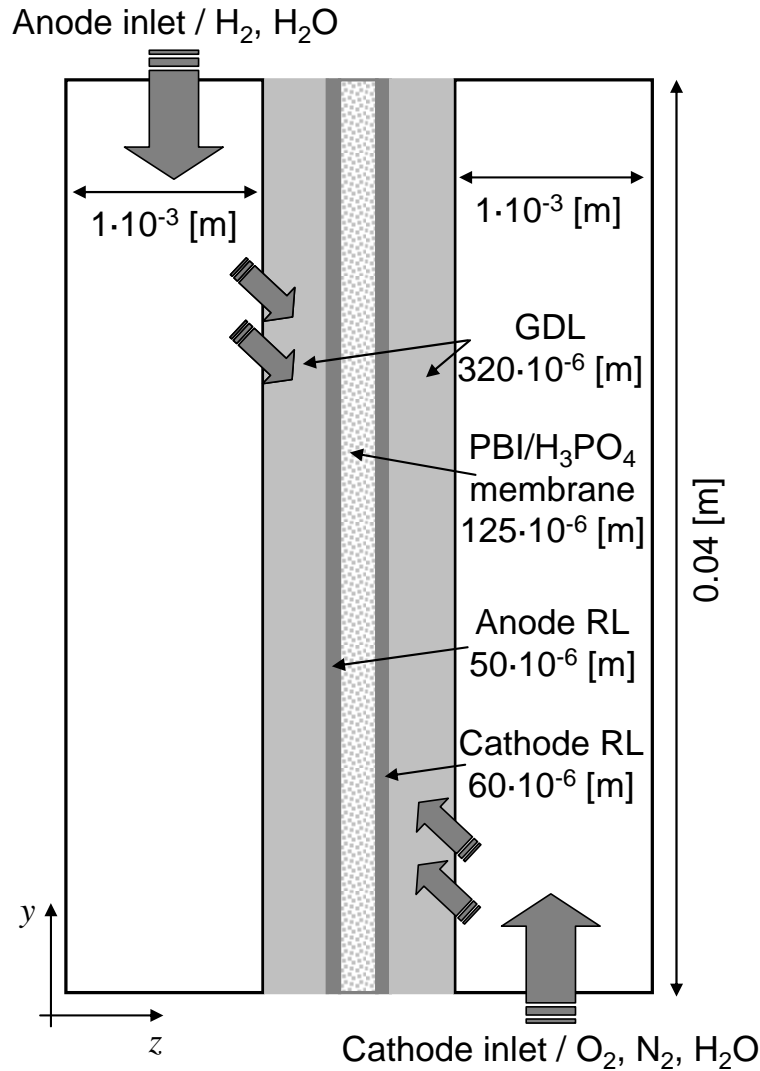
- Reformate gas (higher CO-tolerance)
- No humidification (simplified system design)
- Faster reaction kinetics
- Simplified heat management

Objectives:

- Development a complete large scale 3D HTPEM model (including structural details)
- Model 'validation' with experimental investigations

Challenges:

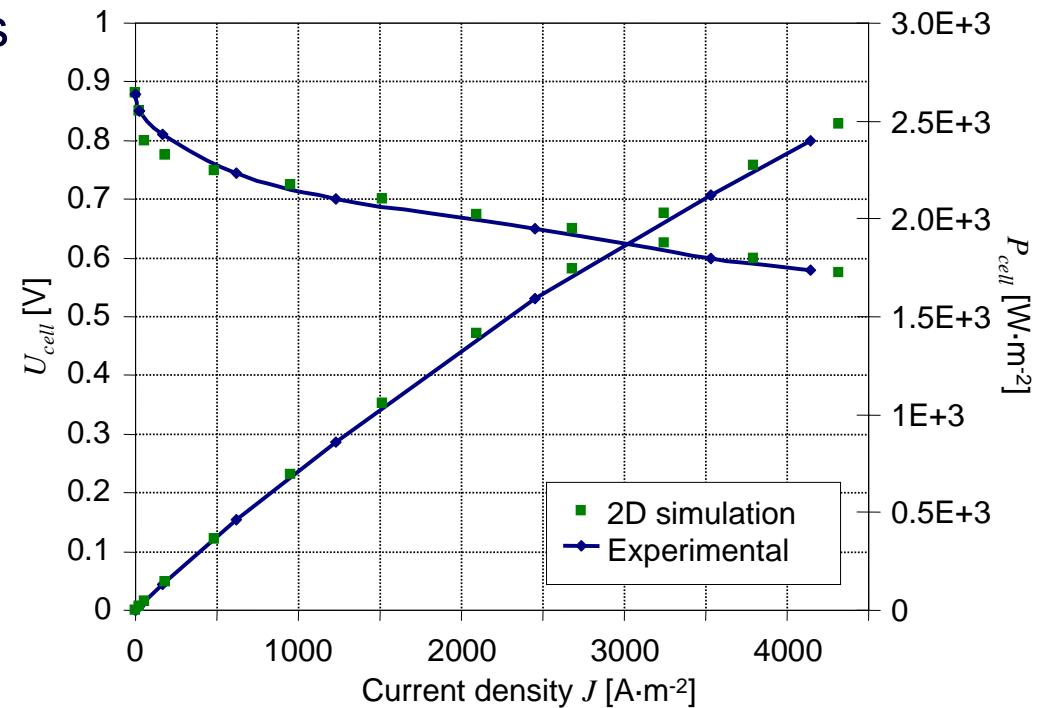
- Only few CFD/FEM models available in open literature (e.g. *Cheddie et al.*, *Peng et al.*, *Korsgaard et al.*)
- (Relatively) new technology (some transport processes not fully understood yet)
- Only few material parameters available



- 2D 'along-the-channel' model
- All conservation laws included
- PBI/H₃PO₄ sol-gel membrane conductivity (empirical equation based on published results)

$$\sigma_m = k_{m,1} - k_{m,2} \cdot T_s^3 + k_{m,3} \cdot T_s^2 - k_{m,4} \cdot T_s$$

- Results

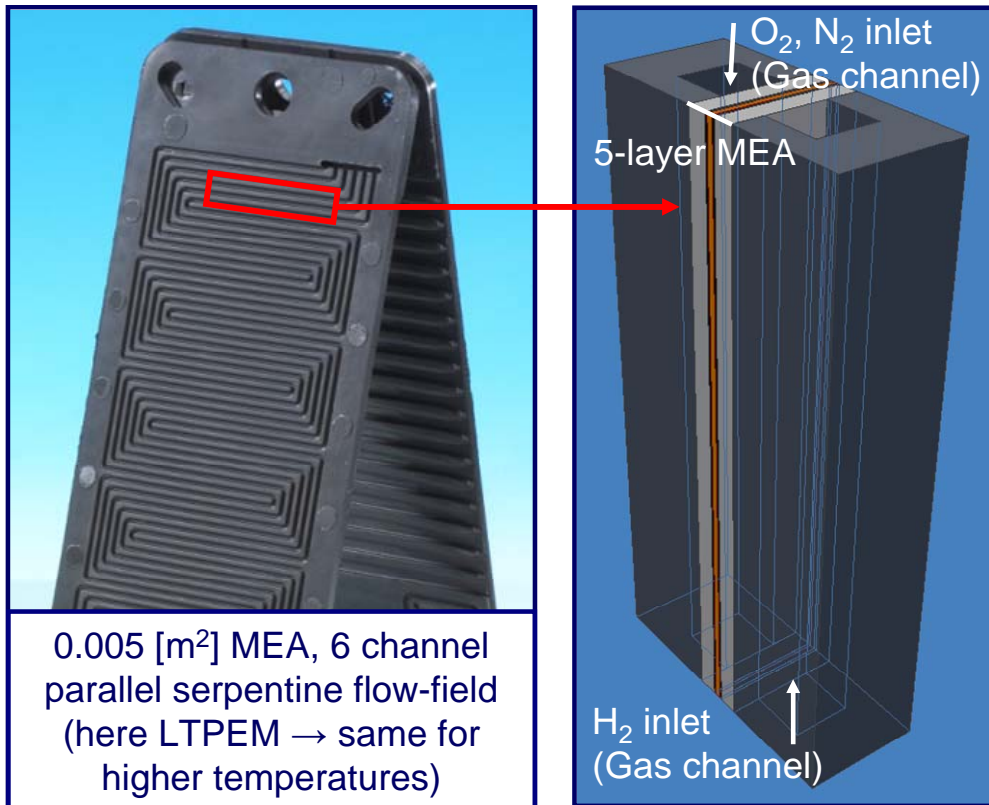


*Siegel, C. et al., Proceedings of the European COMSOL Conference – Vol. 1, Numerical simulation of a high-temperature PEM (HTPEM) fuel cell, 428-434, Petit, J.-M., Squalli, O., Grenoble, France (2007)



- Development of a 3D HTPEM model including gas channels and bipolar-plates
- Enhancement of the already developed and presented 2D study
- All conservation equations (mass, momentum, energy, species, charge) included
- PBI/H₃PO₄ sol-gel membrane conductivity modeled using an Arrhenius equation
- Most physical and material properties updated according the latest publications in the field and personal communications
- More realistic view of HTPEM operation behaviour

3D model → extended to the third dimension (here x)



Dimensions: 0.002 [m] by 0.003 [m]
by 0.01 [m] (x, y, z)

Mass / momentum balance (u, P):

- Gas channel → Navier-Stokes
- Porous media → Brinkmann

Mass / species balance (ω_i):

- Gas channel and porous media → Stefan-Maxwell convection and diffusion

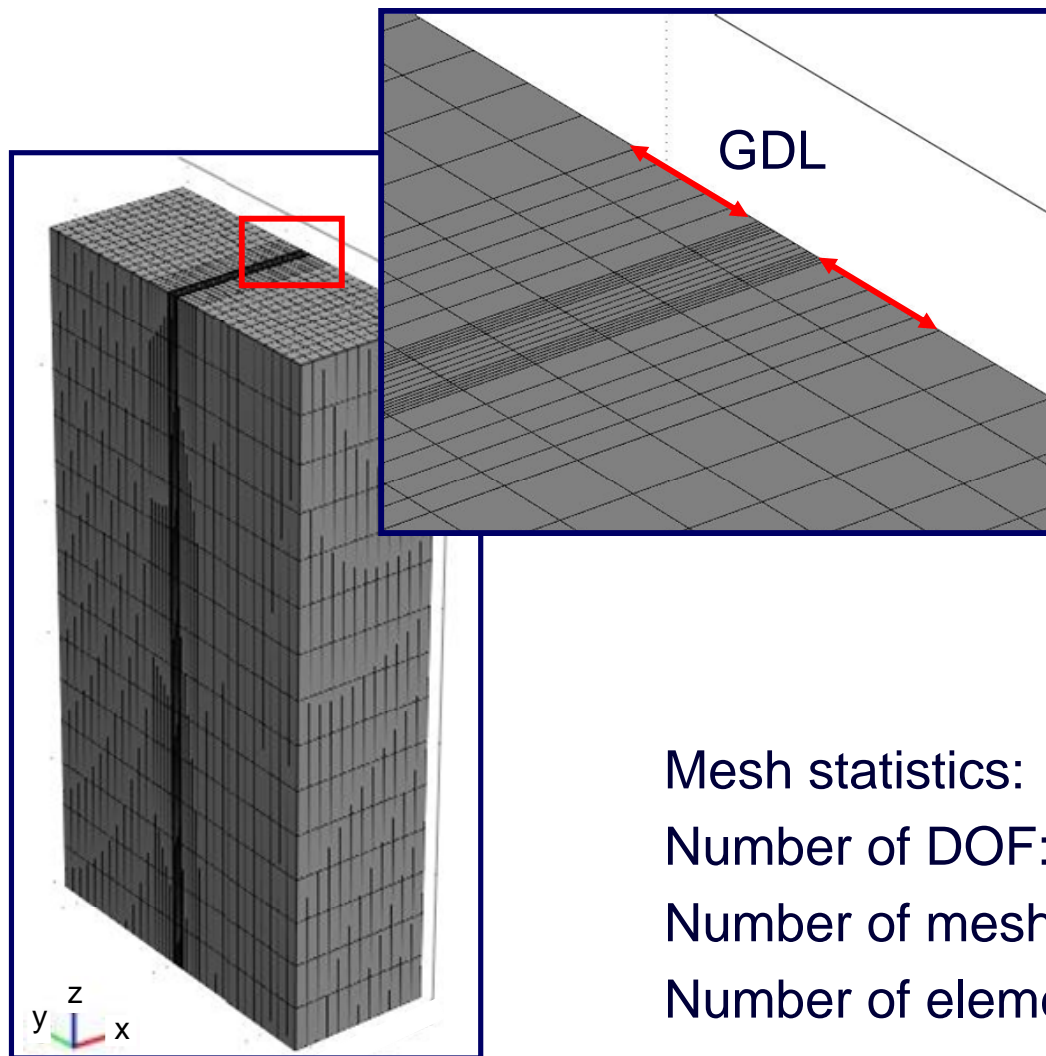
Charge balance (ϕ_i):

- Bipolar-plate, GDL, RL, PBI/H₃PO₄ sol-gel membrane → Conductive media DC

Mass / energy balance (T_i)

- Gas channel, bipolar-plate, GDL, RL, PBI/H₃PO₄ sol-gel membrane → Heat transfer

Boundary condition		Note	Experimental data
Gas flow and pressure	Velocity (inlet) – calculated $u_{i,in} = x_{i,in} \cdot \gamma_i \cdot J \cdot \frac{a_{MEA}}{a_{ch}} \cdot \frac{R \cdot T_f}{n \cdot F} \cdot \frac{1}{P_i}$ $L_e \cdot \nabla_t \cdot (P \cdot I - \eta \cdot (\nabla_t u + (\nabla_t u)^T)) = -\vec{n} \cdot P_{0,e}$ $\nabla_t u = 0$ Pressure (outlet) – measured $\eta \cdot (\nabla_t u + (\nabla_t u)^T) \cdot \vec{n} = 0$ $P = P_{0,out}$	Calculated Measured (ambient pressure)	St. A 1.35 St. C 2.5
Species	Mass fraction (inlet) Convective flux (outlet)	Gas composition Defined	0.25 %RH A 4 %RH C
Charge	Cell voltage (cathode) Reference potential (anode)	Defined	0.95-0.5 [V]
Solid phase temperature	Cell temperature	Defined (controlled)	150-180 [°C]
Fluid phase temperature	Gas temperature (inlet) Convective flux (outlet)	Measured	21 [°C]



Structured mesh to reduce the number of degrees of freedom (DOF)

- GDL 5 elements in y
- RL 5 elements in y
- PBI/H₃PO₄ sol-gel membrane 4 elements in y
- Gas channel 6 x 6 x 14 elements in x , y and z

Mesh statistics:

Number of DOF: 378,139

Number of mesh points: 10,400

Number of elements: 8,820

Minimum element quality: 0.0178

Element volume ratio: 0.06

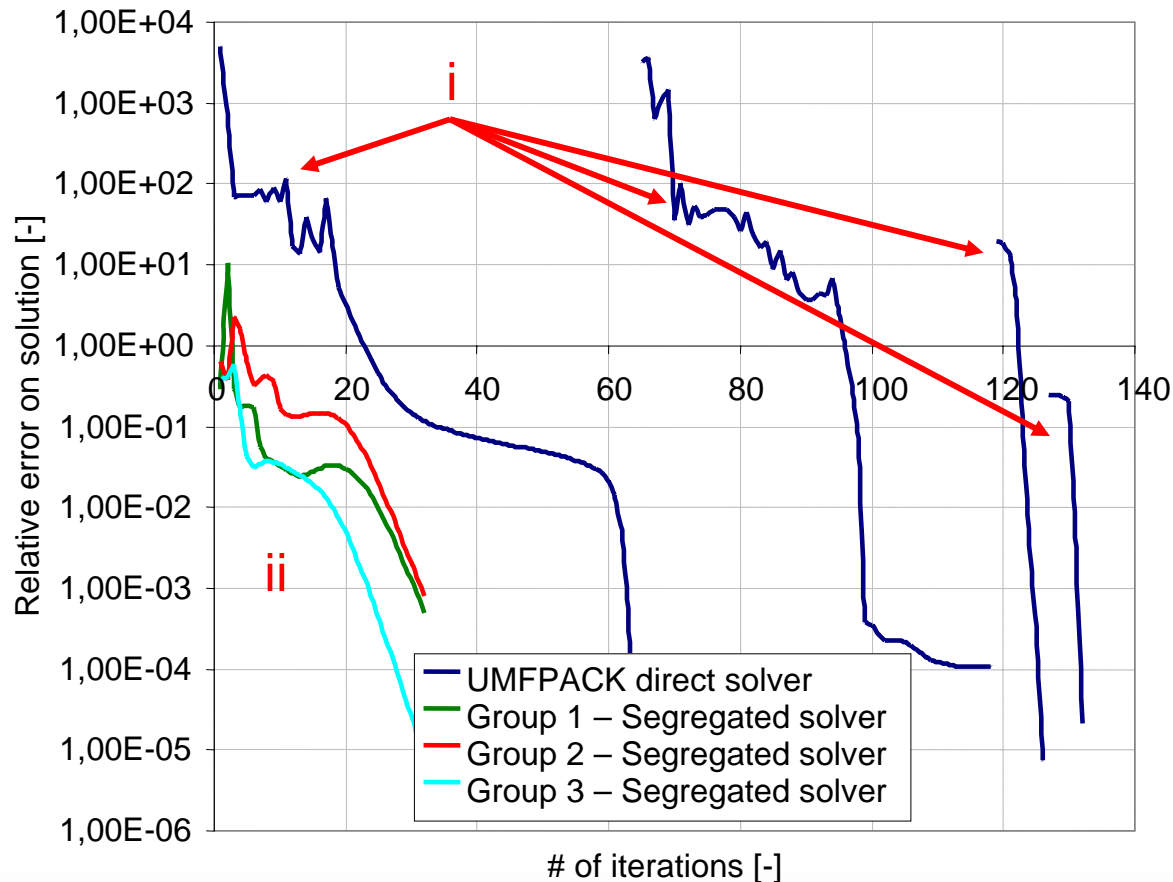
COMSOL MP 3.4 / Quad-core 8GB Ram

i) PARDISO direct solver (± 9000 s)

$$\varphi_i \rightarrow u, P \rightarrow \omega_i \rightarrow T_i$$

ii) Segregated group solver (± 4300 s)

$$u, P \rightarrow \omega_i \rightarrow T_i, \varphi_i$$

**Details:**

Artificial diffusion (streamline)

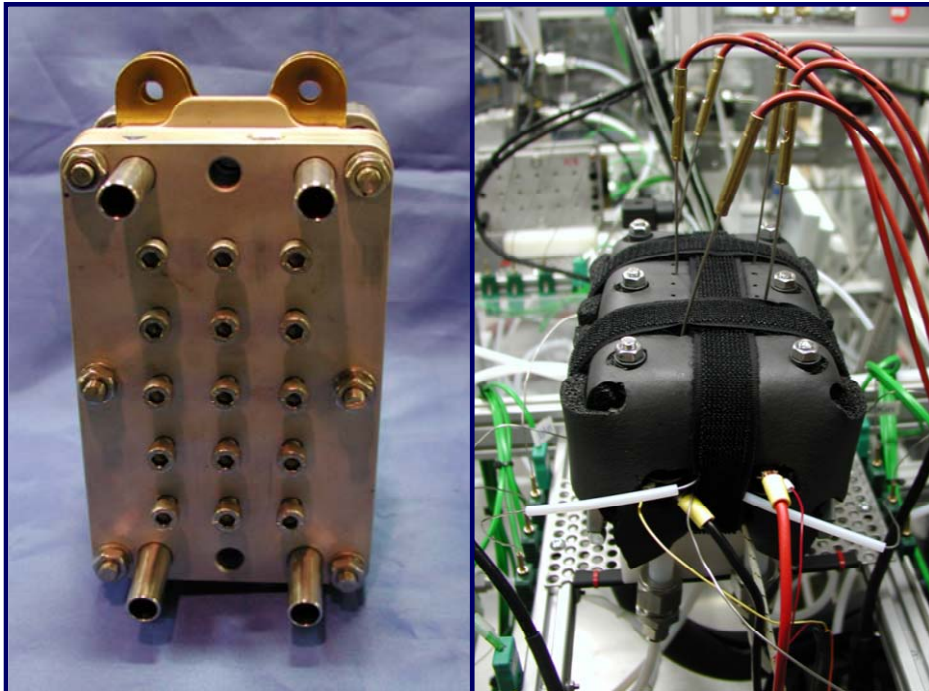
Problem solved sequentially
using solver scripting

Initial conditions updated

Convergence criteria $1 \cdot 10^{-6}$

All tests performed at
the ZBT in Duisburg

Dedicated NI-LabVIEW controlled
teststand for model 'validation'



Design:

- HTPEM single cell assembly (temperature controlled – heated aluminum endplate)

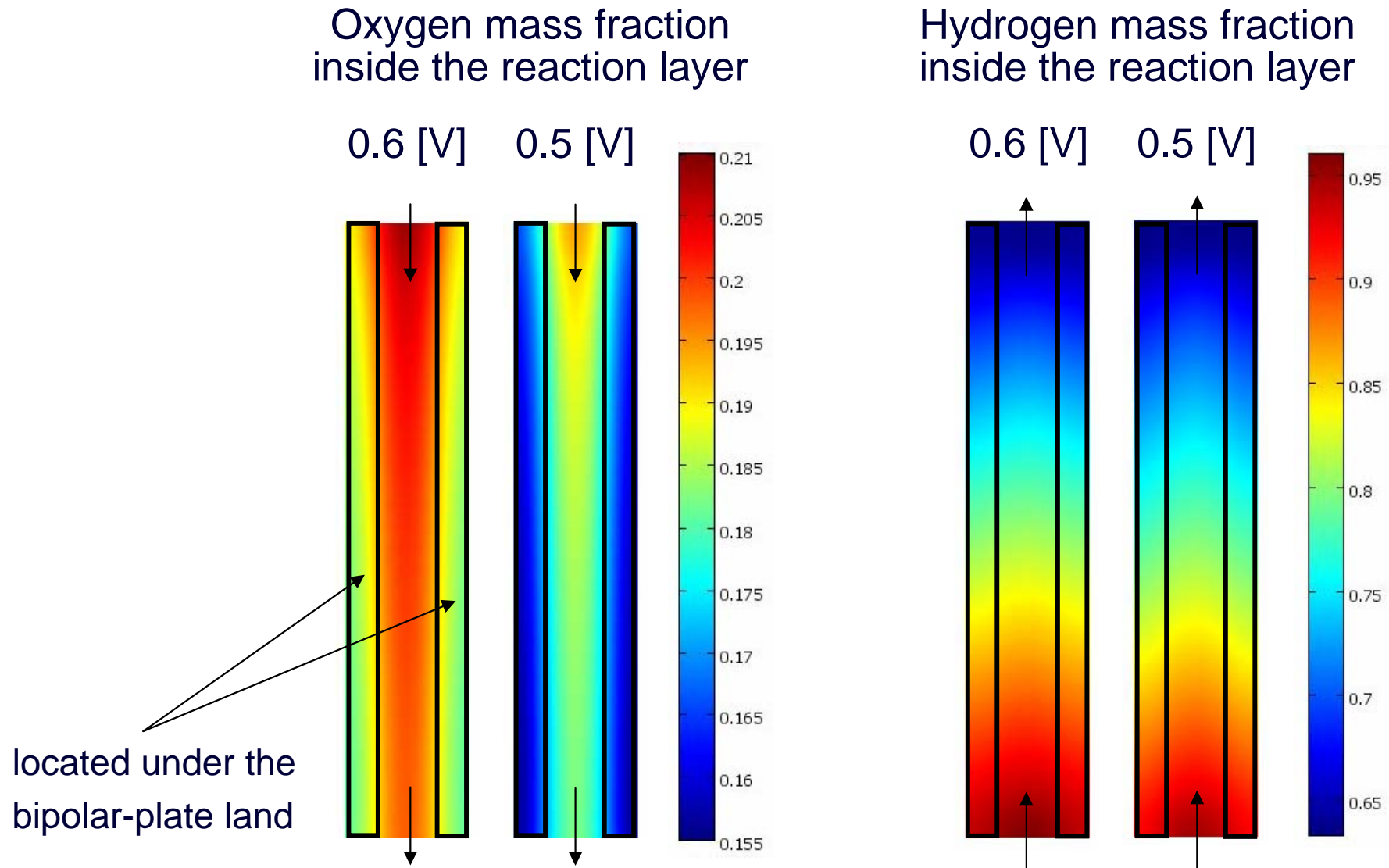
Bipolar-plates:

- Highly conductive (electrical and thermal) inhouse bipolar-plates (polyphenylene sulfide (PPS) based)*
- 6 channel parallel serpentine flow field (anode and cathode side)

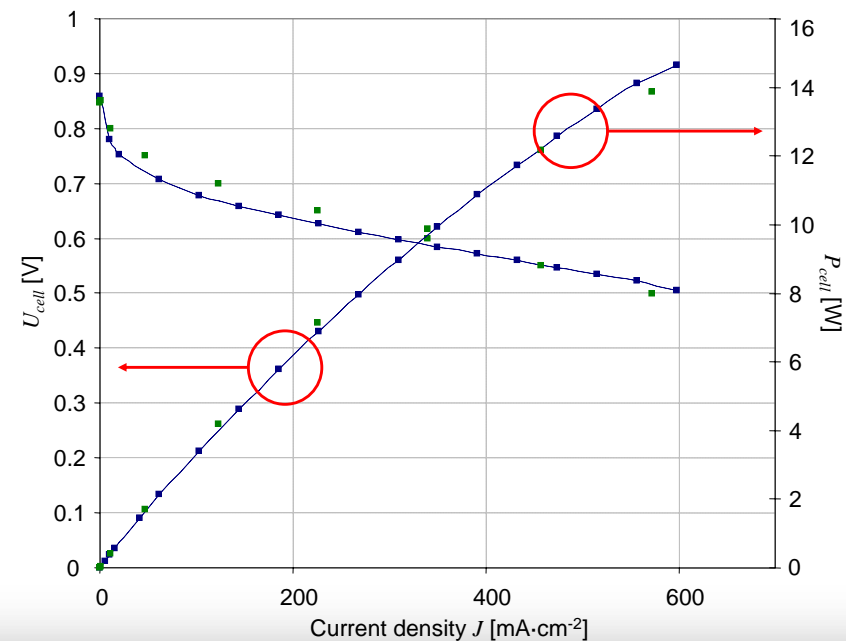
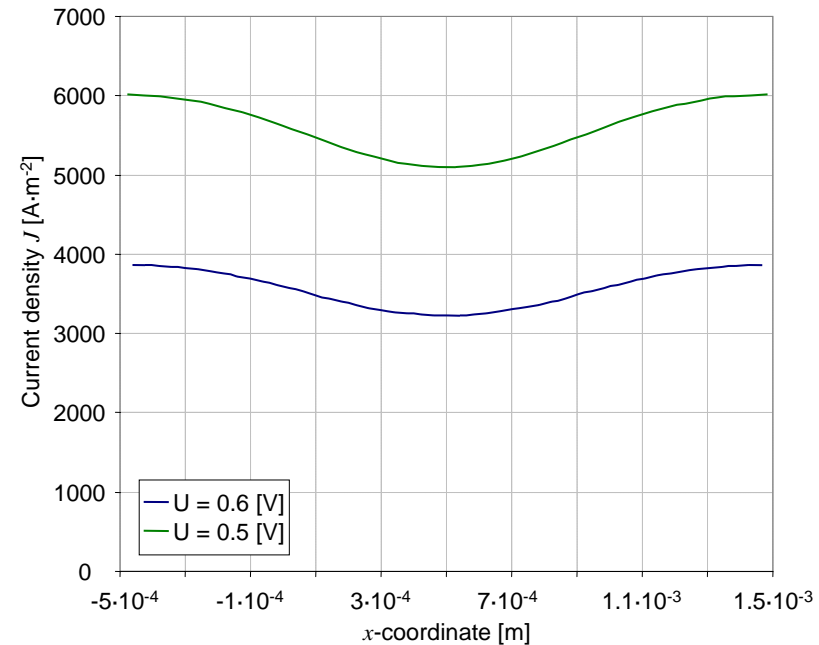
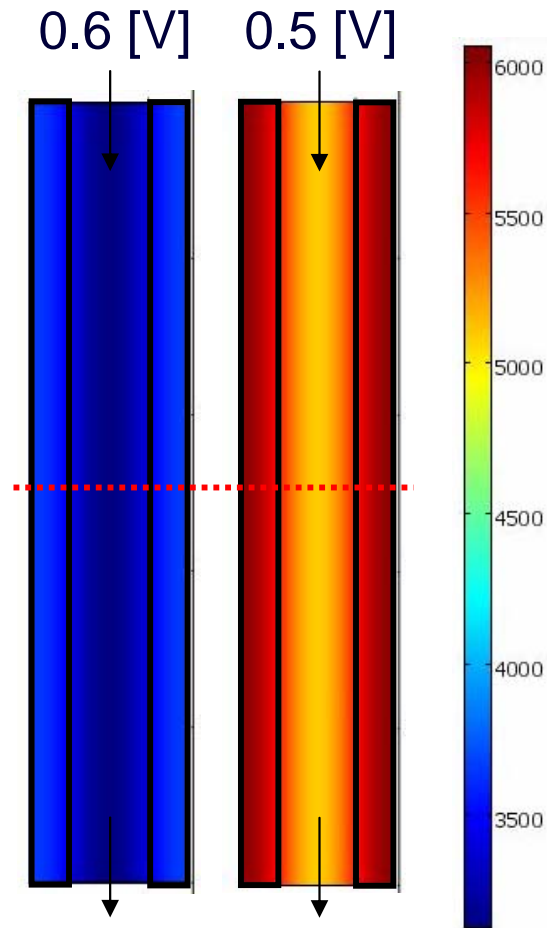
Membrane electrode assembly:

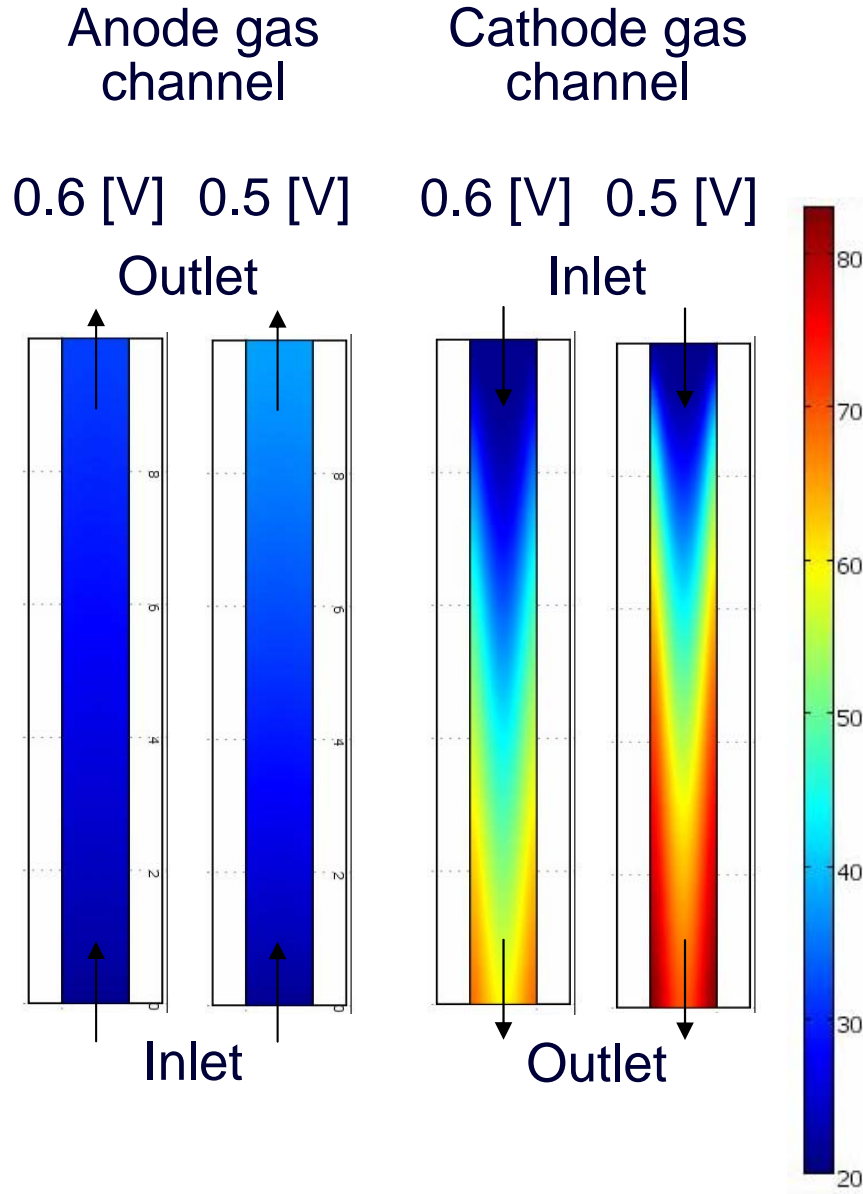
- Highly doped commercially available PBI/H₃PO₄ sol-gel membrane (0.005 [m²])
- Similar to: GDL → E-tek (ELAT)
- Assumed loading 0.01 [kg·m⁻²] (anode) / 0.0075 [kg·m⁻²] (cathode) with a Pt/C-ratio of 0.3 [-]

*Derieth, T. et al., Development of highly filled graphite compounds as bipolar plate material for low and high temperature PEM fuel cells, *J. New Mat. Electrochem. Systems* 11 (2008)

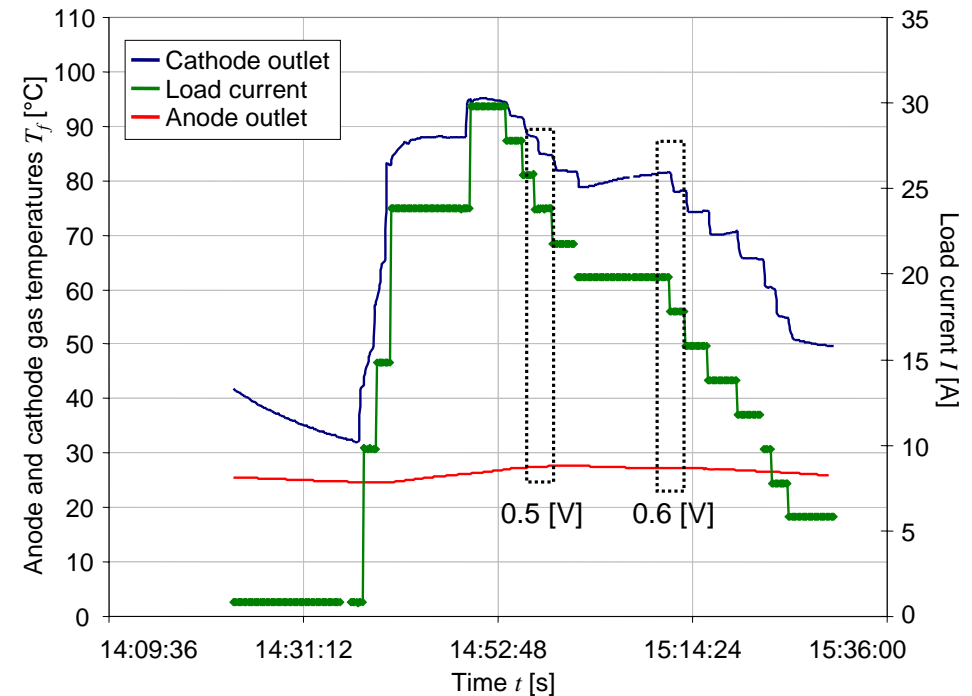


Current density distribution inside the reaction layer (cathode)



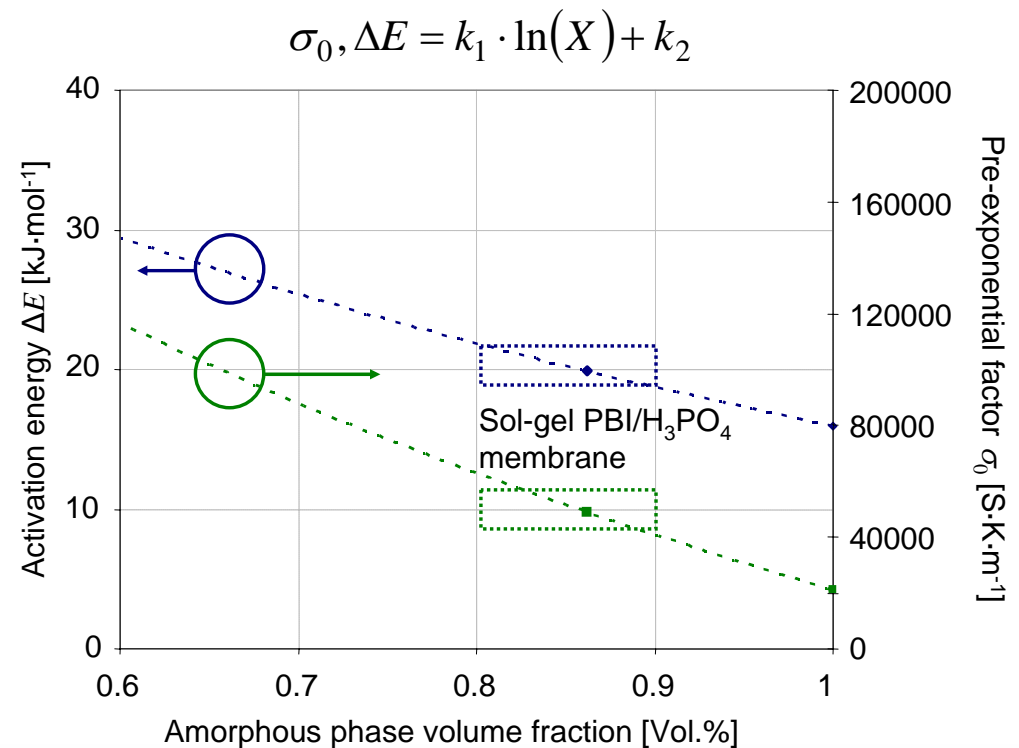
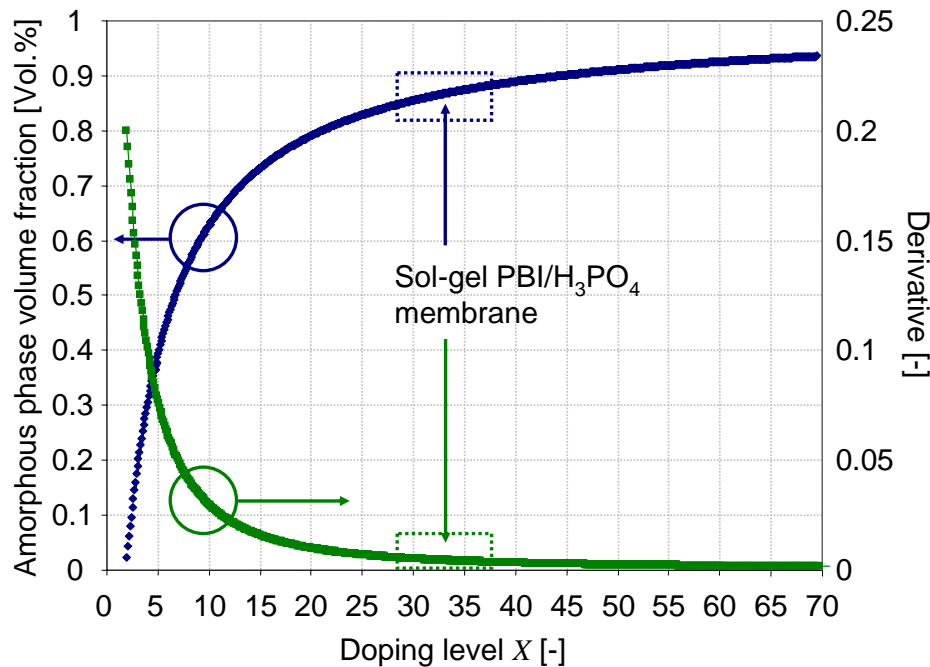


Measured gas outlet temperatures over different load currents



Conductivity dependent on H₃PO₄ doping level

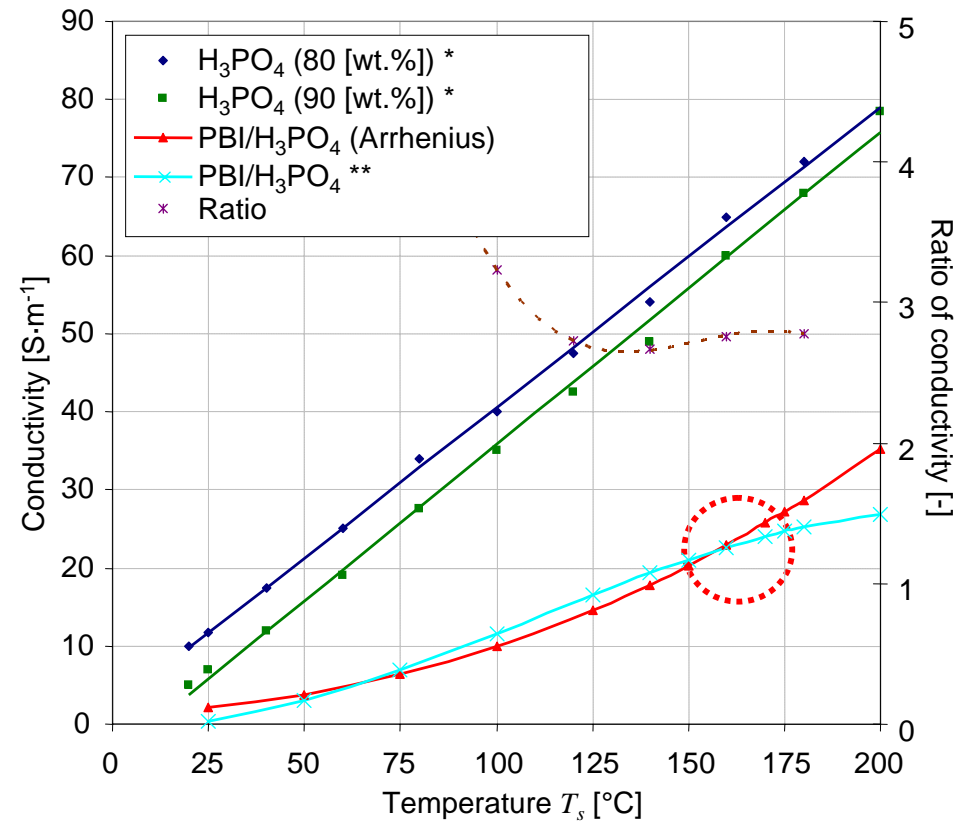
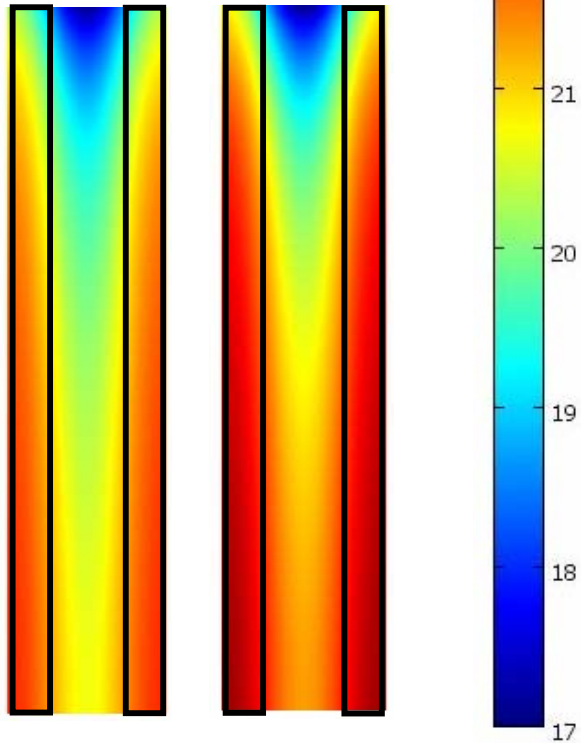
- Number of H₃PO₄ molecules per PBI repeat unit
- Here: Doping level X (and $X-2$) expressed using a density relation (similar to*)
- Amorphous phase volume fraction used to calculate (empirically) the
 - Pre-exponential factor and
 - Activation energy



*Cheddie, D. et al., A two-phase model of an intermediate temperature PEM fuel cell, *Int. J. of Hydrogen Energy* 32 (2007)

PBI/H₃PO₄ sol-gel
membrane conductivity

0.6 [V] 0.6 [V]
160°C 170°C



$$\sigma_0 = \left(\frac{z^2 \cdot F^2}{R} \right) \cdot \alpha \cdot v_0 \cdot d^2 \cdot C \cdot e^{\frac{\Delta S + \Delta S_f}{R}} \quad \sigma = \frac{\sigma_0(k_i, X)}{T_s} \cdot e^{\left(-\frac{\Delta E(k_i, X)}{R \cdot T_s} \right)}$$

$$r_{PBI/PA/H_3PO_4} = \frac{\sigma_{H_3PO_4}(85wt.%) }{\sigma_{PBI/H_3PO_4}} \approx 2.8 - 3.0$$

*Xiao, L. et al., High-temperature polybenzimidazole fuel cell membranes via a sol-gel process, *Chem. Mater.* 17 (2005)

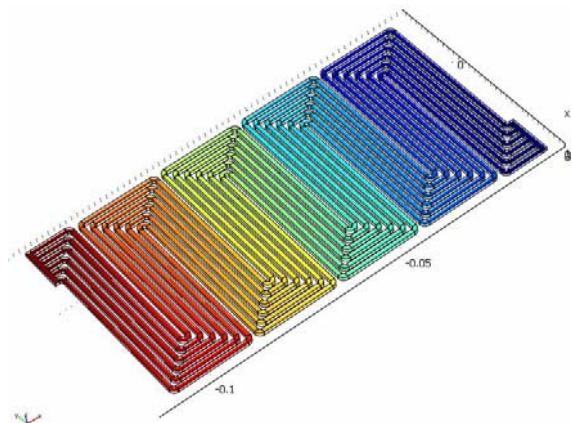
**Chin, D.T. et al., On the conductivity of phosphoric acid electrolyte, *J. Appl. Electrochem.*, 19 (1989)

Development of a 3D HTPEM model

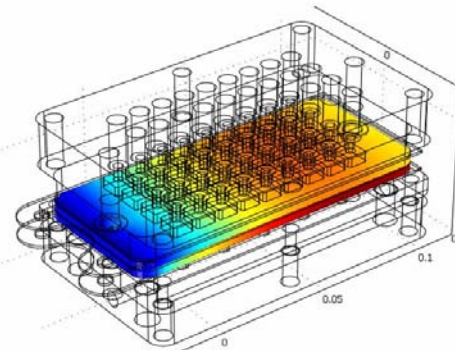
- Ameliorated PBI/H₃PO₄ sol-gel membrane model development and shortly discussed
- Detailed solid- and fluid-phase temperature investigations
- Model not validated but rather compared with experimental data

Further development of a complete HTPEM single fuel cell assembly model by our group

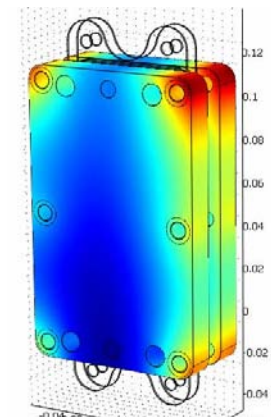
- Update material parameters (continuously)
- Enhance electrochemical part of the model (e.g., agglomerate approach, H₃PO₄ distribution modeling?)



Large scale flow-field simulation
(e.g., pressure losses)



Tempering concepts (e.g. bipolar-
plate temperature distribution)



Structural aspects (e.g.,
overall displacement)

Thank you very much!

Questions, comments? Discussion!