Numerical Investigation of Mass Transfer with Two-Phase Slug Flow in a Capillary

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Outline

- Introduction
- Theoretical Background
- Computation Model
- Results
- Conclusions

Introduction

- Demand in biomedical, chemical reaction engineering, food processing etc.
- Development and application of MEMS technology
- The potential in microfluidic technology
- Experiments with microreactor
- A simple type of microreactor with capillary
- Improve mass transfer for immiscible liquids
- CFD method provides more details

Theoretical Background

Demonstration by using a simple neutralization reaction

$$CH_3COOH + KOH \xrightarrow{k} CH_3COOK + H_2O$$

- Assume it is 2nd order reaction
- Takes place without any additional condition
- Easy to be quantified

Experiment Background

Two phases:

- ❖ Aqueous phase (250 mol/m³ KOH aqueous solution)
- ❖ Organic phase (650mol/m³ acetic acid mixed with kerosene)



Demonstration of Two-Phase slug flow with chemical reaction

Theoretical Background

- Assume: rate constant k=0.001m³mol⁻¹s⁻¹
- Kinetic equations

$$R_{AA} = \frac{d[CH_{3}COOH]}{dt} = -k[CH_{3}COOH][KOH]$$

$$R_{KOH} = \frac{d[KOH]}{dt} = -k[CH_{3}COOH][KOH]$$

Reaction rate

$$R = -k[CH_3COOH][KOH]$$

Theoretical Background

Mass transfer by convection-diffusion

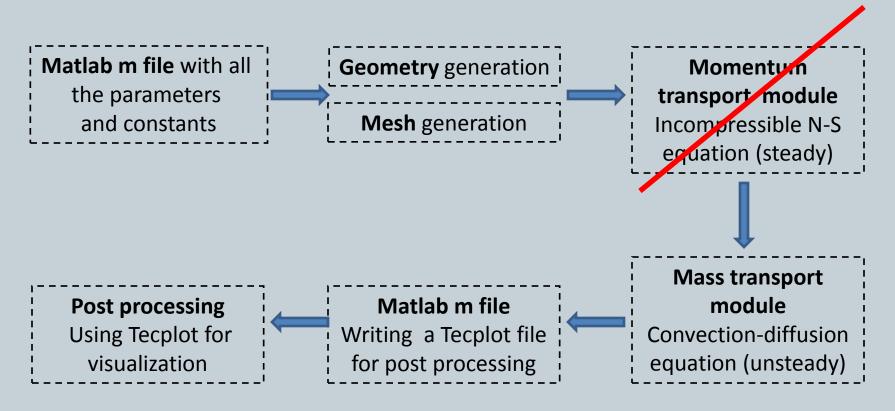
$$\frac{\partial C_{mn}}{\partial t} + \vec{u} \cdot \nabla C_{mn} - \nabla \cdot (D_{mn} \nabla C_{mn}) - R = 0$$

Mass transfer by diffusion

$$\frac{\partial C_{mn}}{\partial t} - \nabla \cdot (D_{mn} \nabla C_{mn}) - R = 0$$

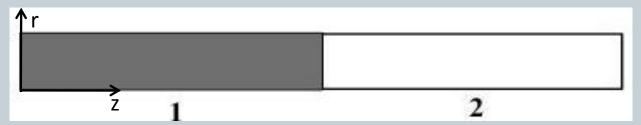
Where,
$$D_{mn} = 7.4 \times 10^{-8} \frac{T \sqrt{\varphi_n M_n}}{\mu_n (M_m/\rho_m)^{0.6}}$$
 (Wilke-Chang correlation)

Computation Scheme



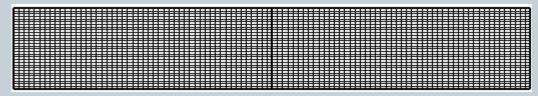
Computation Model

- 2-D Axisymmetric
- Two axes are independent
- Computational domain



Domain 1 represents aqueous phase; Domain 2 represents organic phase

Mesh



Structured mesh

❖ Element number: 1500

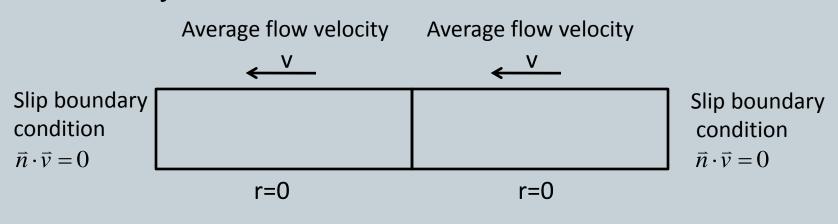
❖ Node number : 3131 2010 COMSOL Conference, Boston, MA

Convection-diffusion case

Solver

- COMSOL Multiphysics with Matlab package
- Discretize governing equation with Finite Element Method
- ❖ Assumption: neglecting gravitational force and surface tension effect
- Solve steady incompressible Navier-Stokes equation
- Solve convection-diffusion equation with implement of flow field

Boundary condition for flow field



Convection-diffusion case

- Boundary condition for solving mass transport
 - Solve mass transport for two species separately
 - ❖ Assume no KOH is transported into organic phase

$$\begin{bmatrix} 2 \\ 4 \end{bmatrix} C_{KOH} = 0, \nabla C_{kOH} = 0$$

- $\vec{n} \cdot \nabla C_{KOH} = 0$ was applied to the boundary 1, 2, 3 and 4
- Solve mass transport of AA in two phases

	2	5	
1			6
	r=0	r=0	

- Periodic boundary condition applied to 1 and 6
- $\vec{n} \cdot \nabla C_{AA} = 0$ was applied to the boundary 2 and 5

Pure diffusion case

Solver

- Solve diffusion equation
- COMSOL Multiphysics with Matlab package
- Discretize governing equation with Finite Element Method
- Solve mass transport for two species separately

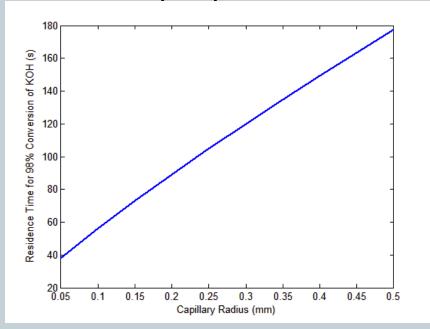
Boundary condition

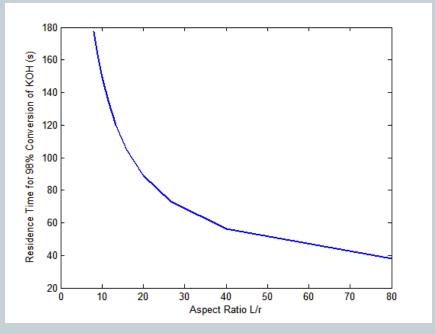
- Periodic boundary condition at interface
- No mass flux through the capillary wall
- ❖ Holds r=0 at central axis

Results (Pure diffusion case)

- Slug length: 4mm; Capillary radius (50µm-500µm)
- Residence time to achieve 98% conversion of KOH

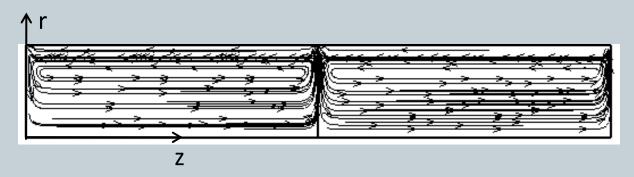
 Aspect ratio is the ratio of slug length to capillary radius (L/r)





Results (Convection-diffusion)

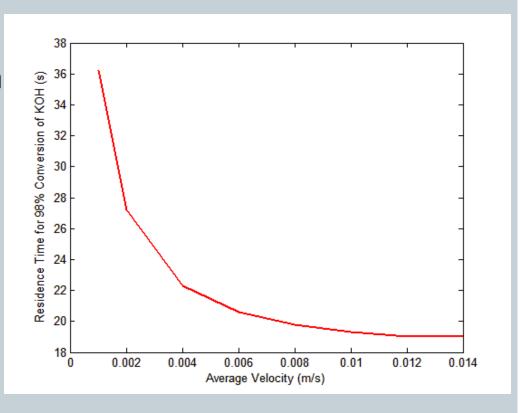
Streamline plot



- Internal circulation formed inside each slug
- Mass transported not only by diffusion but also by convection
- ❖ Increases effective interfacial area
- Increase concentration gradient

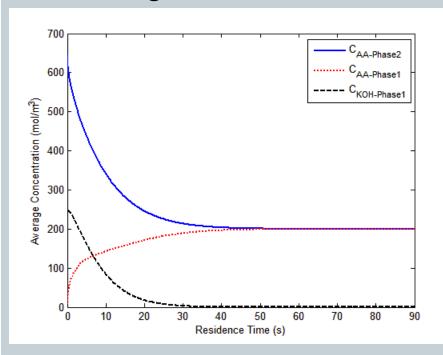
Results (Convection-diffusion)

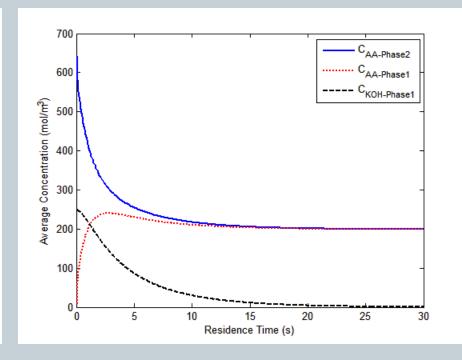
- Capillary radius: 250µm
- Slug length: 4mm
- L/r=16
- Residence time to achieve 98% conversion of KOH



Results (Convection-diffusion)

Average concentration varies with time





The variation of average concentration of two species at **v=0.002m/s** with capillary radius of 250 μm

The variation of average concentration of two species at **v=0.014m/s** with capillary radius of 250 μm

Concentration distribution

Capillary radius of 250 μm

• $C_{KOHmax} = 250 \text{ mol/m}^3$

at t=0s

• C_{KOHmax} =42.99 mol/m³ at u=0.002m/s at t=15s

• $C_{KOHmax} = 14.15 \text{ mol/m}^3$ at u = 0.008 m/s at t = 15 s

• $C_{KOHmax} = 12.14 \text{ mol/m}^3$ at u = 0.014 m/s at t = 15 s

The concentration distribution and maximum concentration value of KOH during the titration for different velocity at residence time **t=15s**

Concentration distribution

- Average velocity: 0.002m/s
 Slug length: 4mm
- Capillary radius: 250 µm
- $C_{AAmin} = 0 \text{mol/m}^3$

 C_{AAmax} =650mol/m³ at t=0s

- $C_{AAmin} = 100.32 \text{mol/m}^3$
- $C_{AAmax} = 387.46 \text{ mol/m}^3 \text{ at t} = 10.3 \text{ s}$

- $C_{AAmin} = 171.19 \text{ mol/m}^3$
- $C_{AAmax} = 235.59 \text{ mol/m}^3 \text{ at t} = 25.2 \text{ s}$

The concentration distribution of AA in two phases

Concentration distribution

- Average velocity: 0.014m/s
 Slug length: 4mm
- Capillary radius: 250 µm
- $C_{AAmin} = 0 \text{mol/m}^3$

 C_{AAmax} =650mol/m³ at t=0s

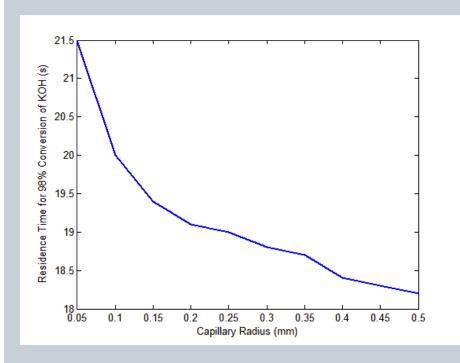
- $C_{AAmin} = 207.89 \text{ mol/m}^3$
- $C_{AAmax} = 217.77 \text{ mol/m}^3 \text{ at t} = 10.3 \text{ s}$

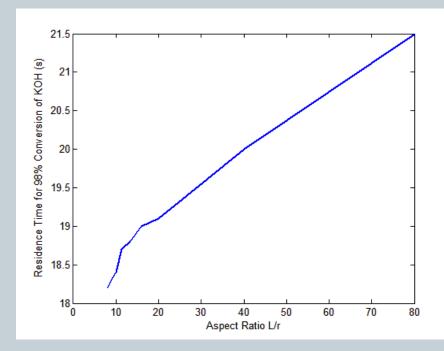
- $C_{AAmin} = 199.21 \text{ mol/m}^3$
- $C_{AAmax} = 199.67 \text{ mol/m}^3 \text{ at t} = 25.2 \text{ s}$

The concentration distribution of AA in two phases

Results (Convection-Diffusion)

- Average flow velocity :0.014m/s
- Slug length: 4mm
- Residence time to achieve 98% conversion of KOH





Conclusions

- COMSOL Multiphysics with Matlab package allows for rapid analysis
- Average flow velocity, slug size and capillary size all have effect on mixing efficiency
- Residence time decreases with increasing average flow velocity for a fixed aspect ratio
- Residence time increases with increasing the aspect ratio for a fixed average flow velocity
- Two-phase slug flow has potential benefit on immiscible fluids mixing

Acknowledge

Dr. Brian H. Dennis

Thank you for your attention!

Question?